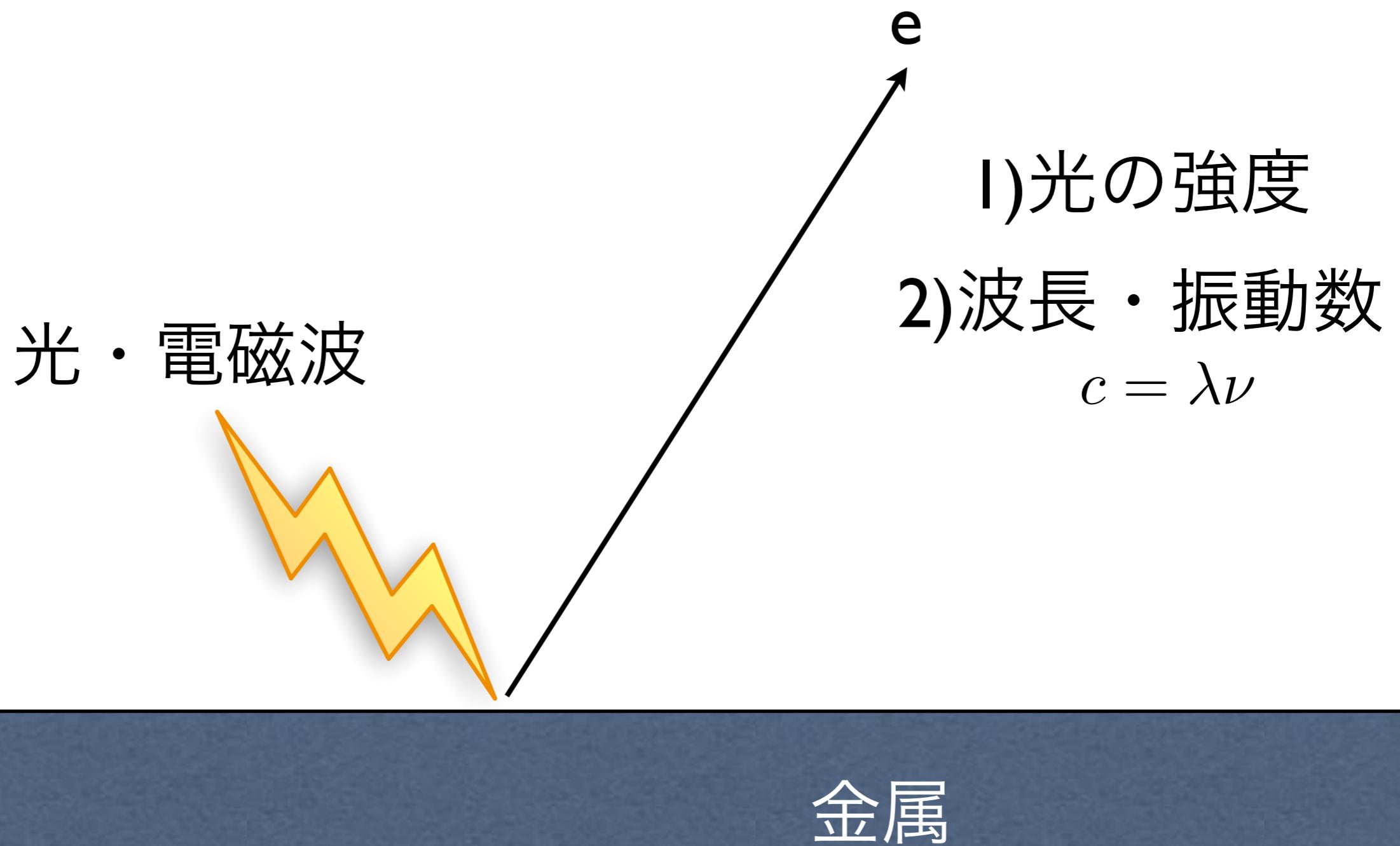
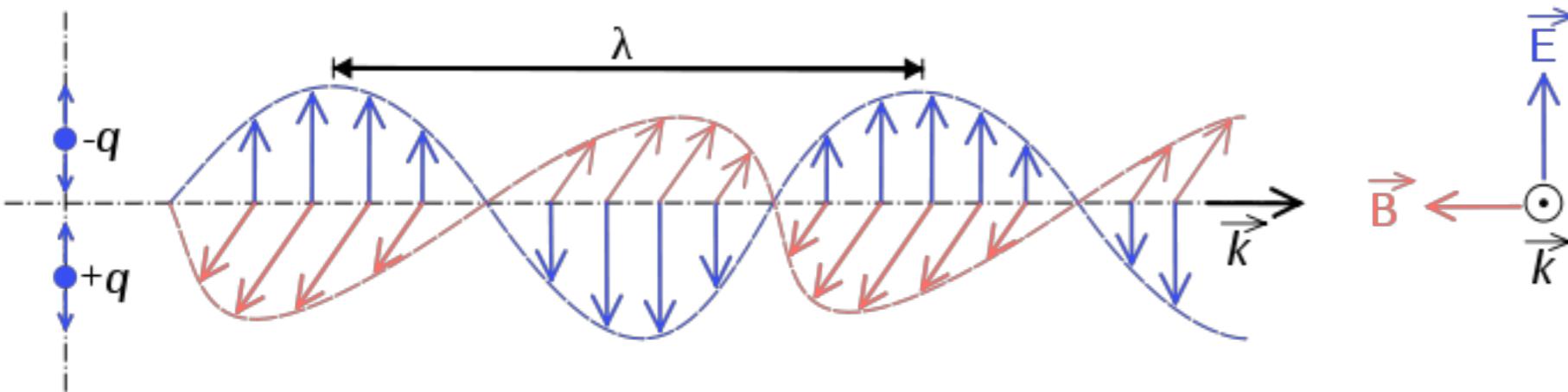


光電効果

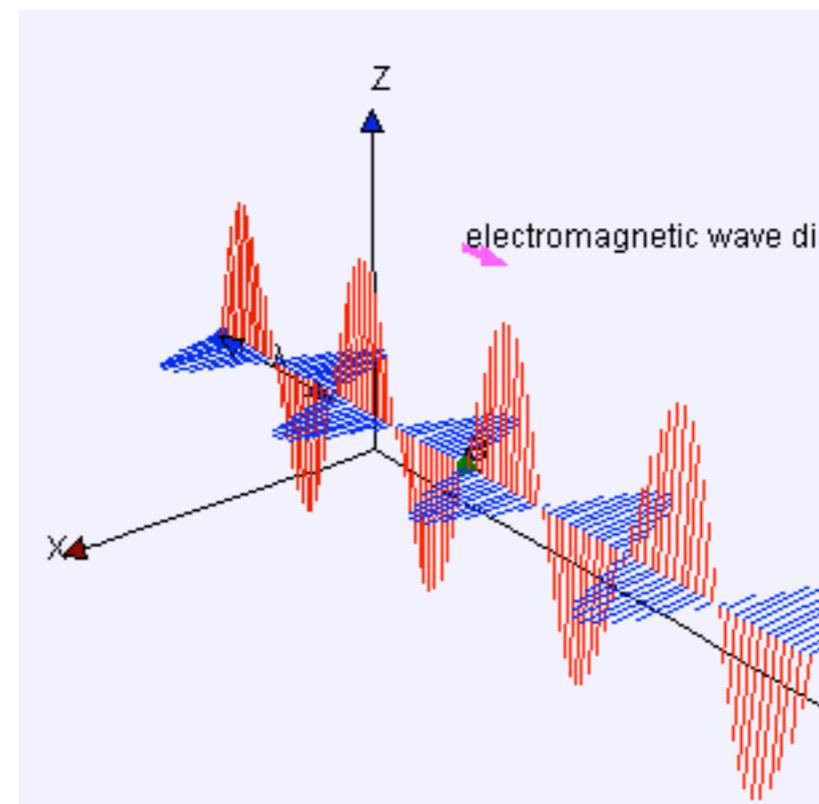




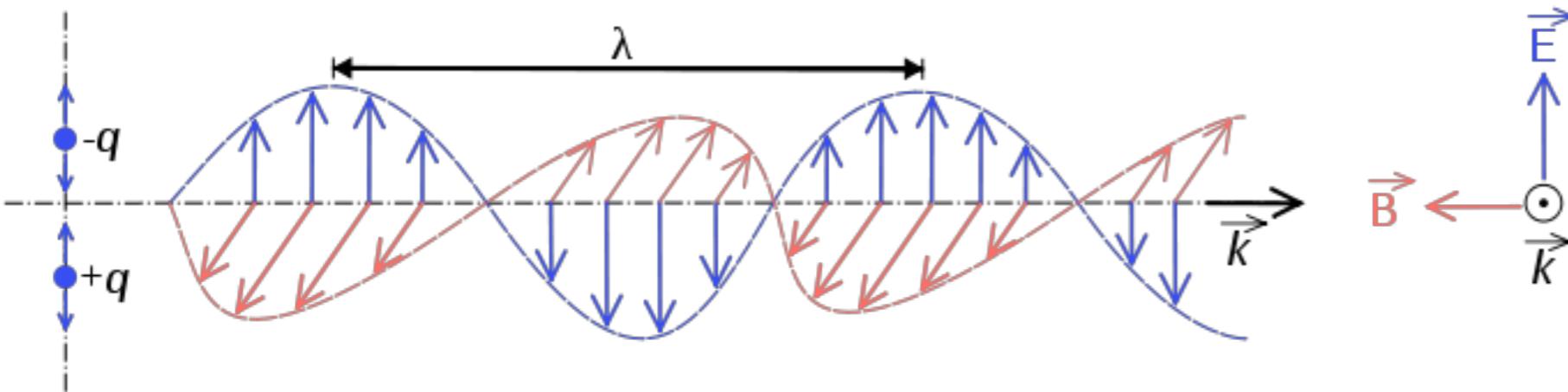
電磁波のエネルギー (古典)

$$\begin{aligned}\epsilon_0 \mathbf{E}^2 &= \epsilon_0 E_0^2 \cos^2(kx - \omega t) \\ &= \epsilon_0 E_0^2 \cos^2[2\pi(\frac{x}{\lambda} - \nu t)]\end{aligned}$$

$$c = \lambda\nu$$



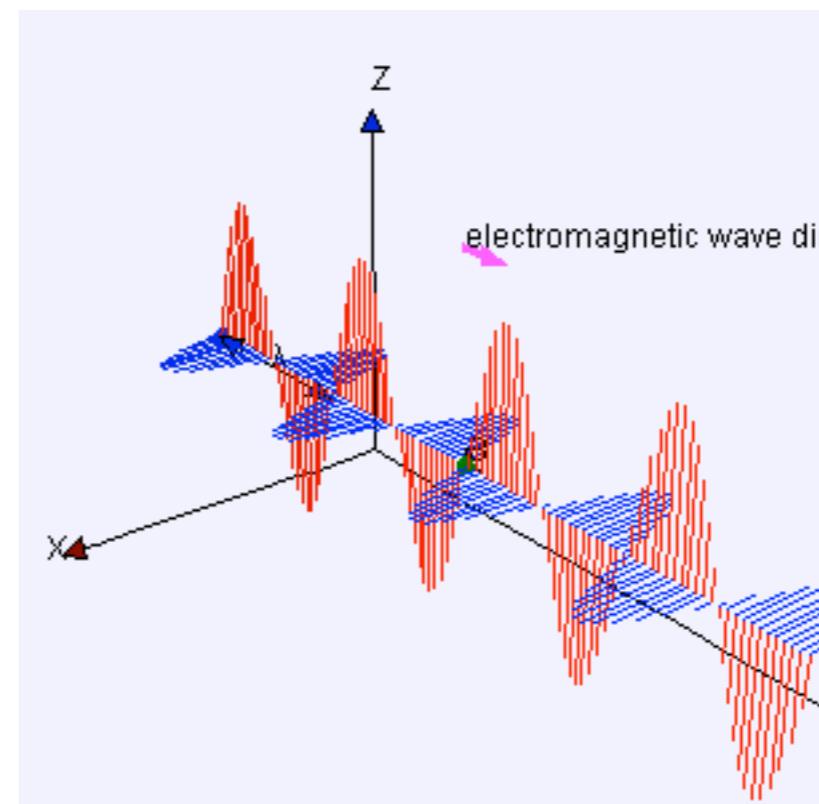
光の振幅（明るさ）を増していくばエネルギーは増加



電磁波のエネルギー (古典)

$$\begin{aligned}\epsilon_0 \mathbf{E}^2 &= \epsilon_0 E_0^2 \cos^2(kx - \omega t) \\ &= \epsilon_0 E_0^2 \cos^2[2\pi(\frac{x}{\lambda} - \nu t)]\end{aligned}$$

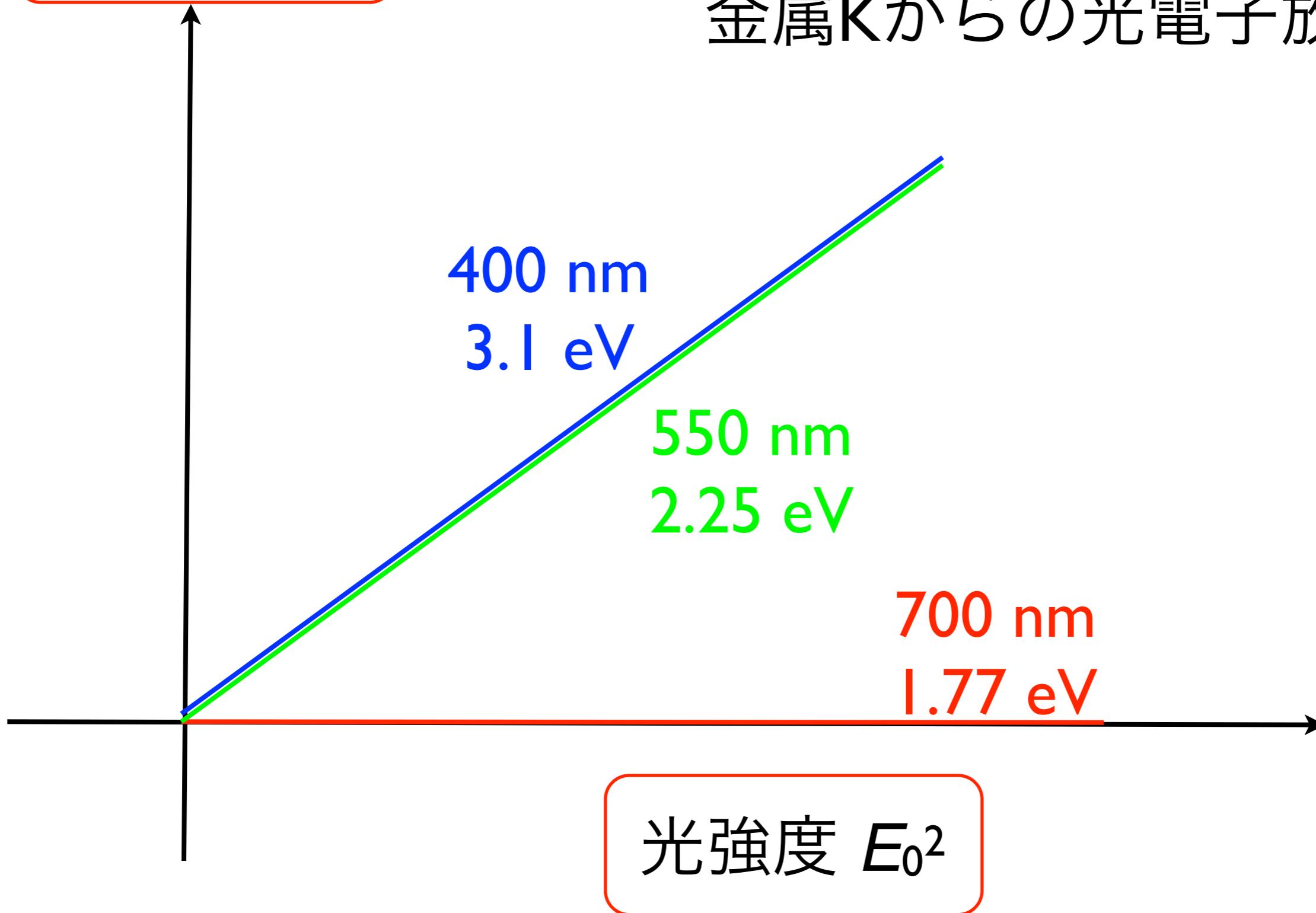
$$c = \lambda\nu$$

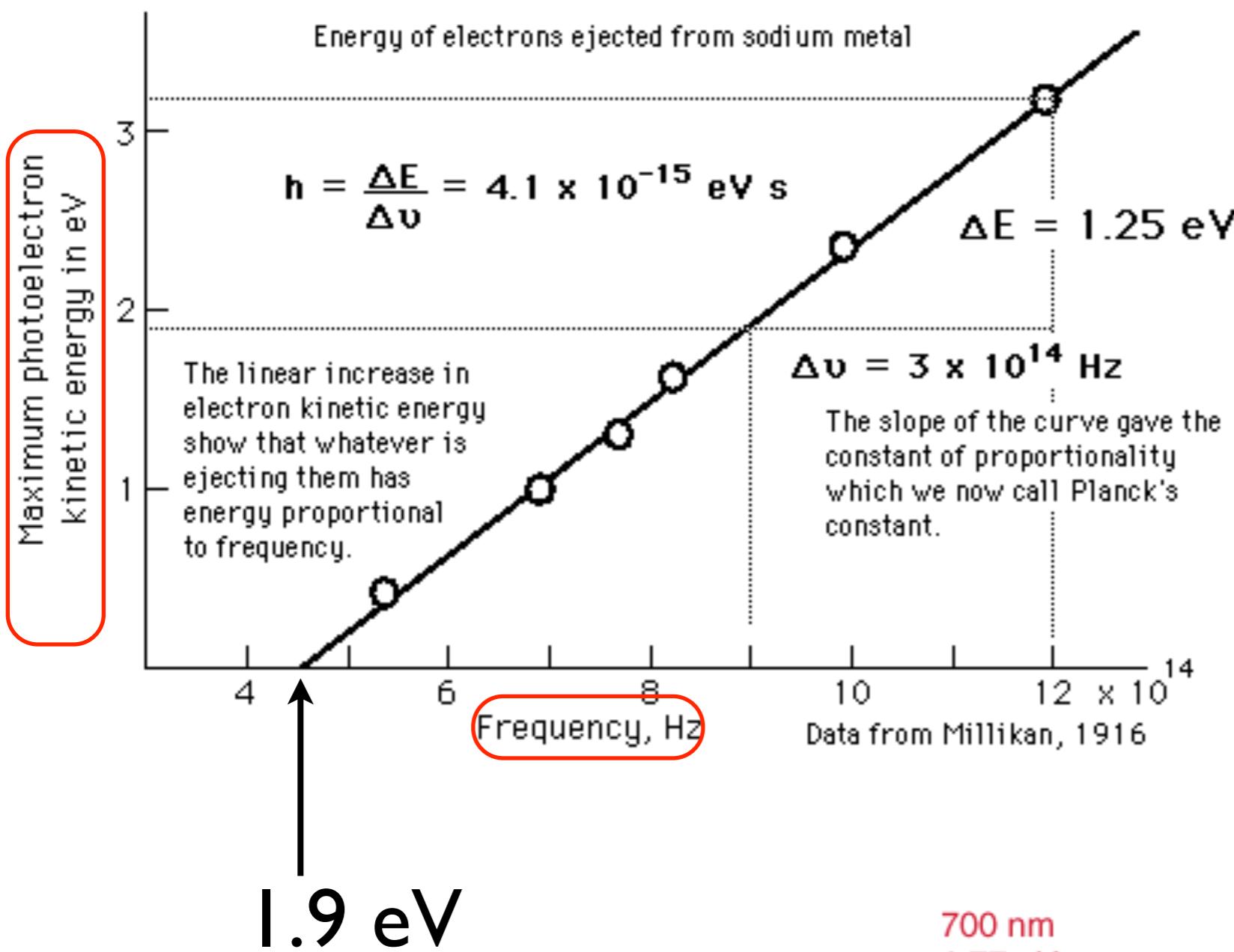


光の振幅（明るさ）を増していくばエネルギーは増加

光電子の数

金属Kからの光電子放出

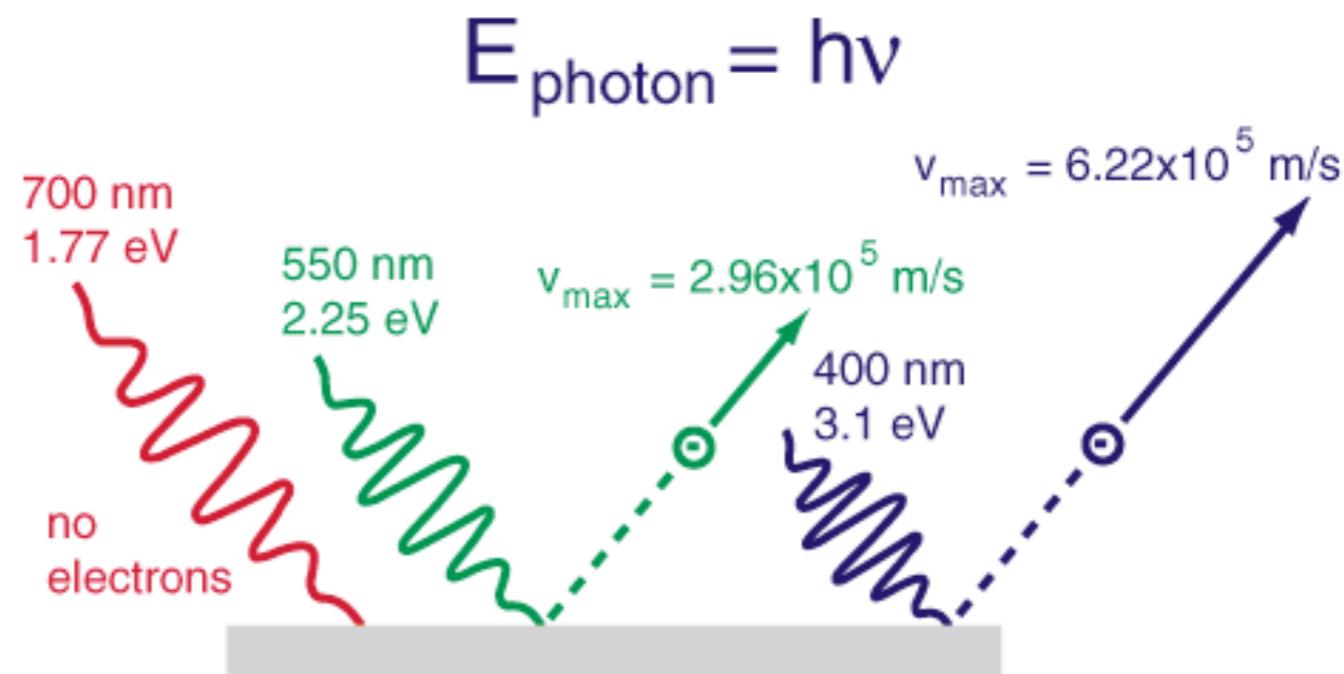




$$KE = \frac{1}{2}mv^2 = h\nu - \phi$$

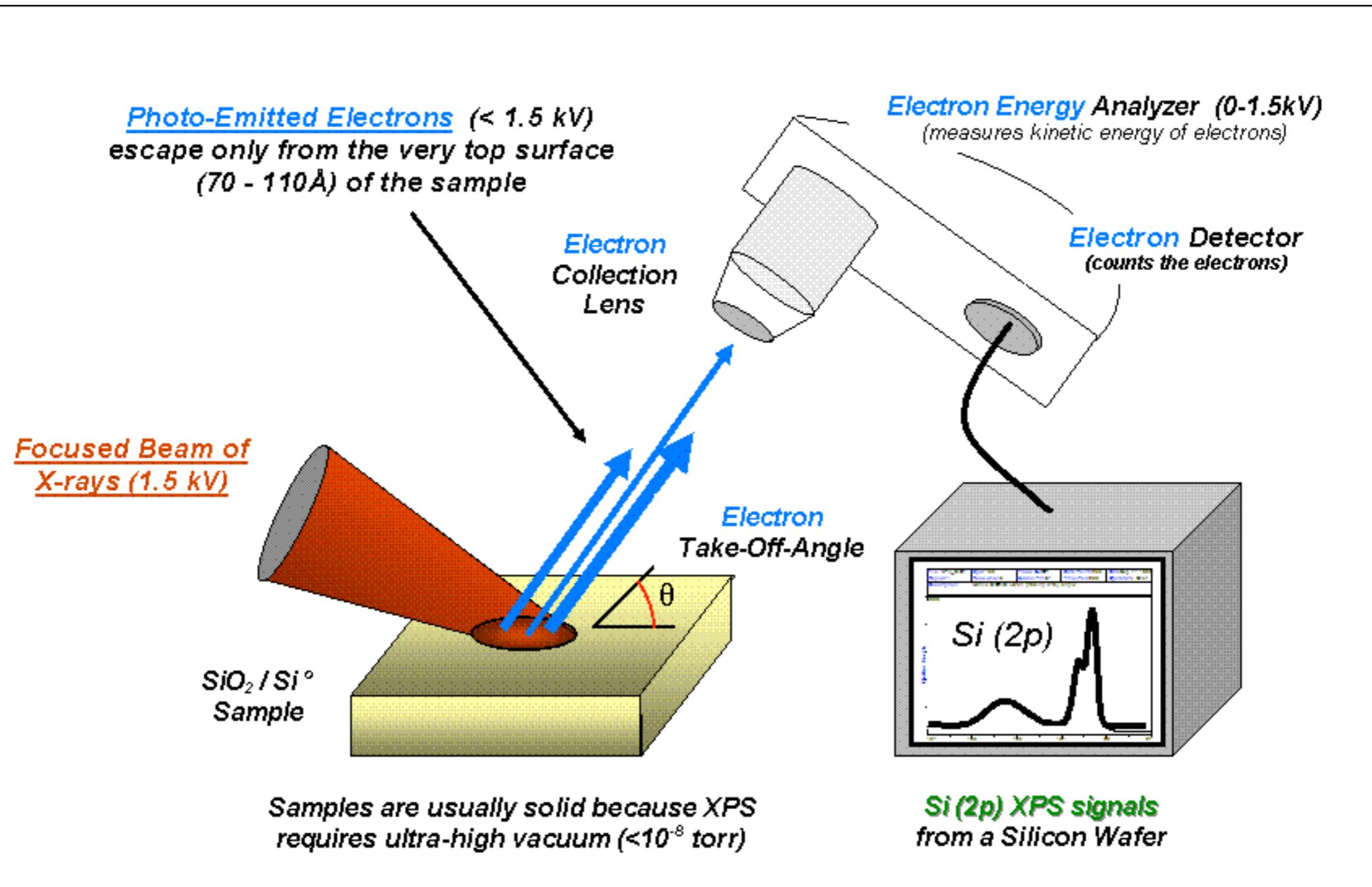
$$h\nu_0 = \phi$$

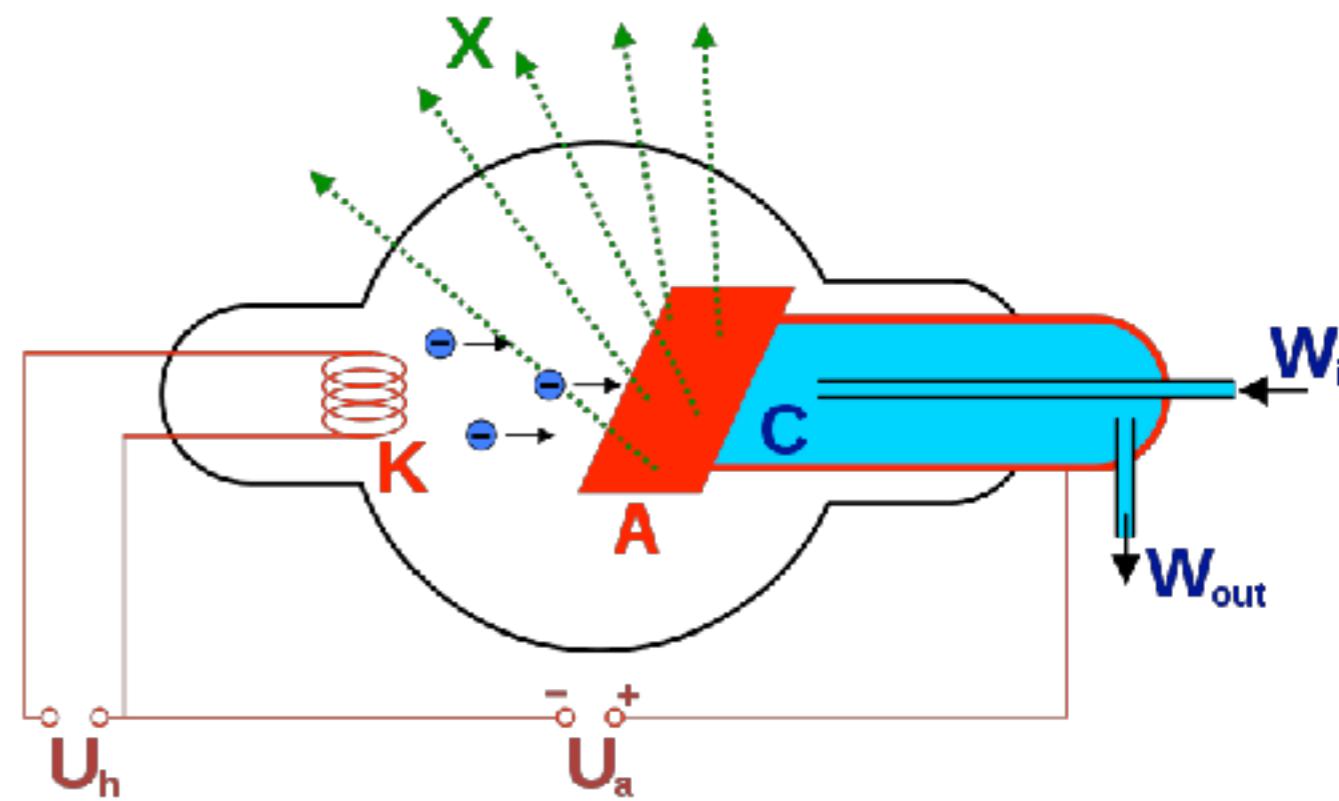
$$h = 4.135667513 \times 10^{-15} \text{ eV s}$$



Photoelectric effect

X線光電子分光 (X-ray photoelectron spectroscopy)





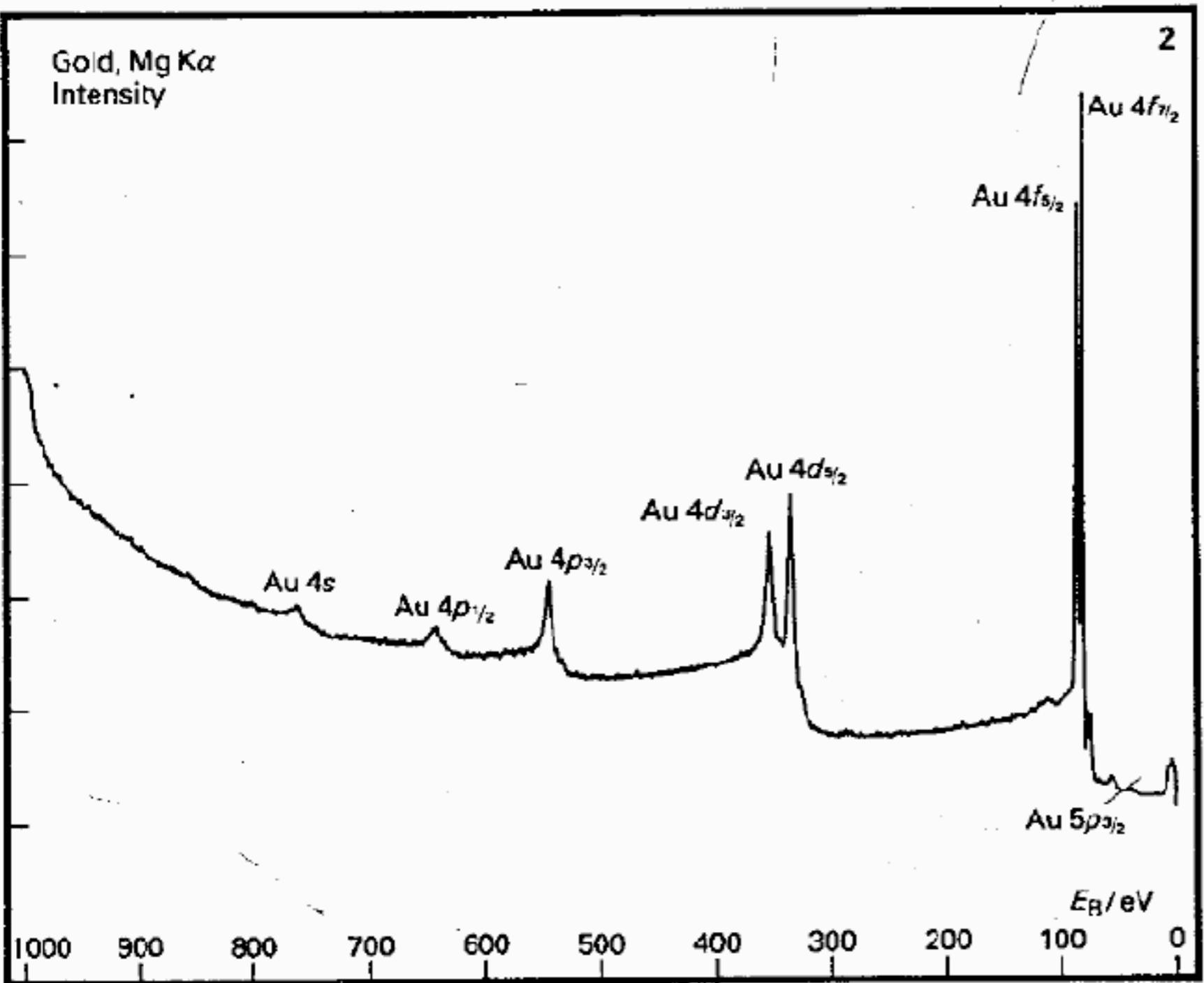
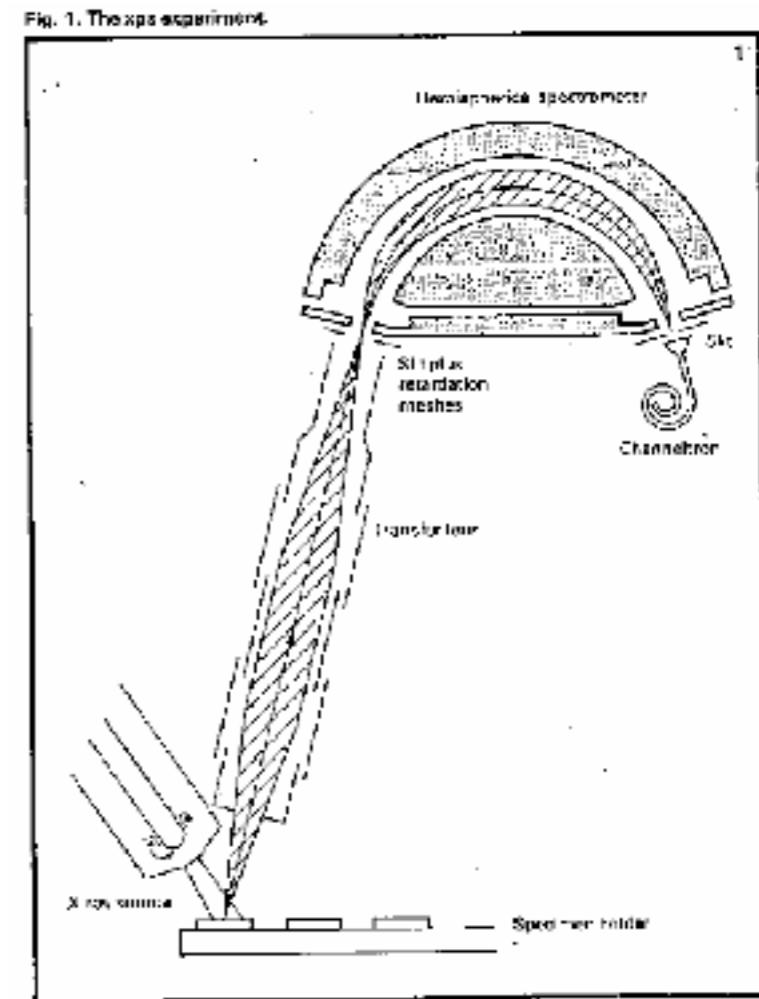


Fig. 2. The gold xps spectrum using Mg K α radiation.³



Au
 4p splitting 96.40 eV
 4d splitting 18.10 eV
 4f splitting 3.67 eV

XPS

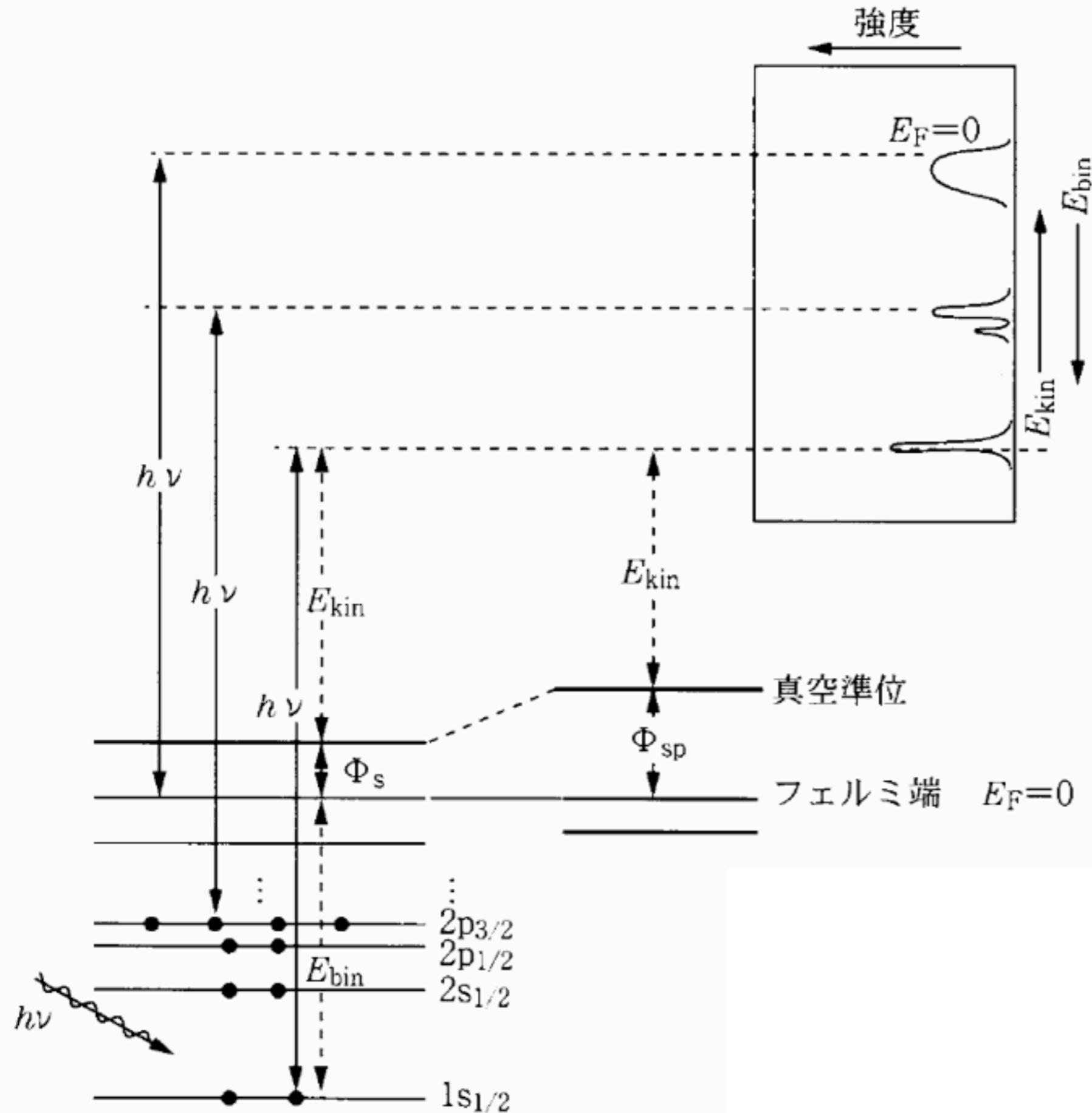


図 2-15 XPS の原理

XPS

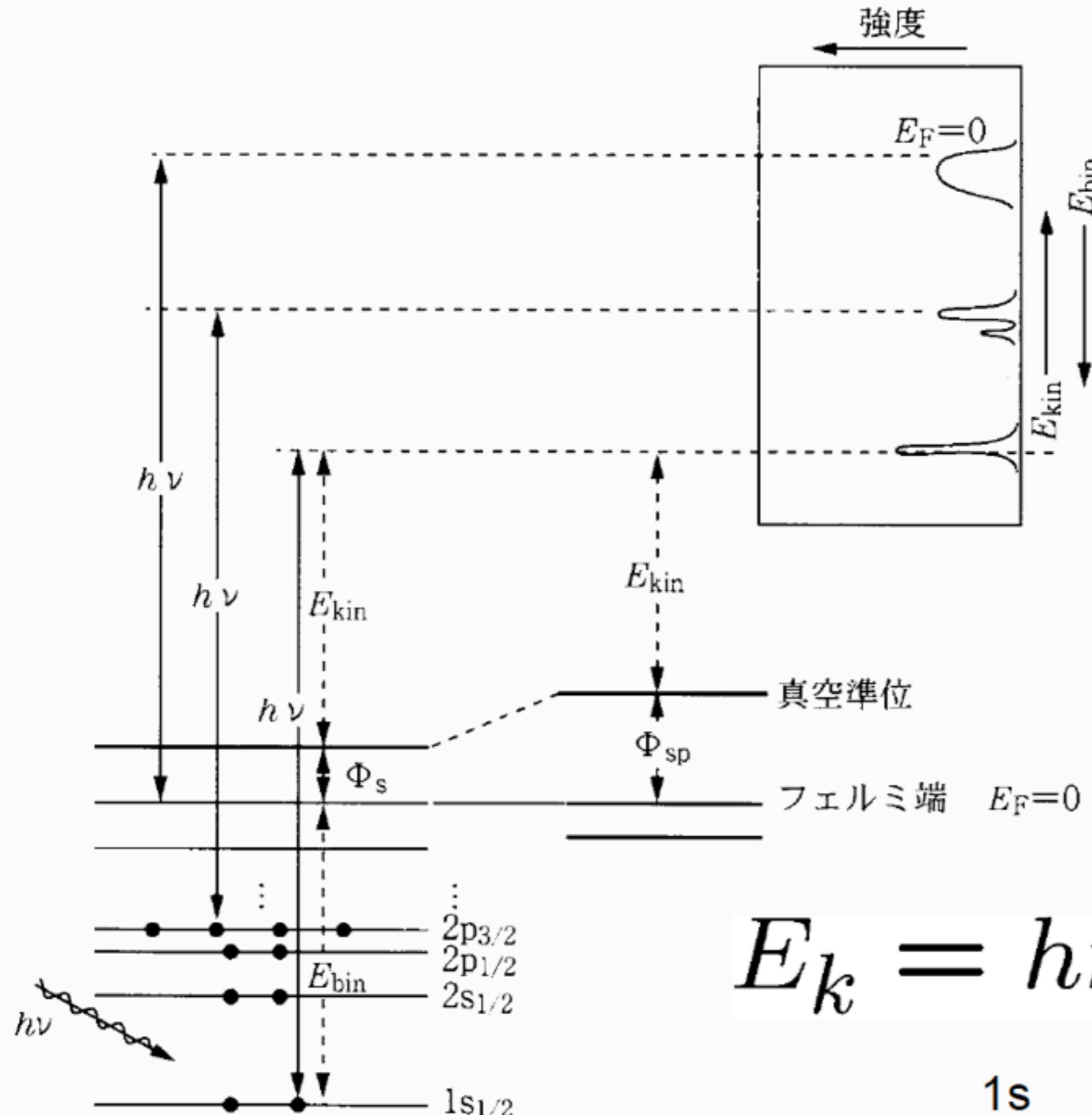


図 2-15 XPS の原理

n, l, m : 1s, (2s, 2p), (3s, 3p, 3d), (4s, 4p, 4d, 4f), ...

縮退 2 2 6 2 6 10 2 6 10 14

$2(2l+1)$

スピン軌道相互作用による微細構造

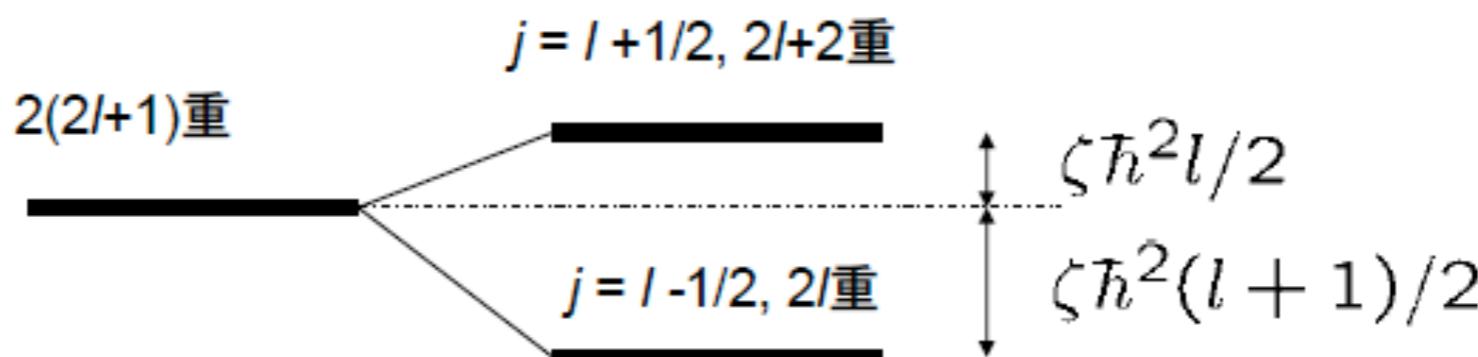
$$H' = \zeta \mathbf{l} \cdot \mathbf{s}$$

主量子数 n , 方位量子数 l , 内量子数 j

nl_j

1s_{1/2} 2s_{1/2} 2p_{1/2}(2) 2p_{3/2}(4)

3s_{1/2} 3p_{1/2} 3p_{3/2} 3d_{3/2}(4) 3d_{5/2}(6)



$l=0, j=1/2$

$s: j = s_{1/2}$

縮退 2

$l=1, j=1/2, 3/2$

$p: j=p_{1/2}, j=p_{3/2}$

2

$l=2, j=3/2, 5/2$

$d: d_{3/2}, d_{5/2}$

4

4

6

	$l=0, j=1/2$	$l=1, j=1/2, 3/2$	$l=2, j=3/2, 5/2$
縮退	$s: j = s_{1/2}$	$p : j=p_{1/2}, j=p_{3/2}$	$d: d_{3/2}, d_{5/2}$
	2	2	4

Au

4p splitting 96.40 eV
 4d splitting 18.10 eV
 4f splitting 3.67 eV