

Fundamental Physical Constants — Frequently used constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_{r}
speed of light in vacuum	c	299 792 458	m s^{-1}	exact
Newtonian constant of gravitation	G	$6.674\,30(15) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	2.2×10^{-5}
Planck constant*	h	$6.626\,070\,15 \times 10^{-34}$	J Hz^{-1}	exact
	$\hbar$	$1.054\,571\,817 \dots \times 10^{-34}$	J s	exact
elementary charge	e	$1.602\,176\,634 \times 10^{-19}$	C	exact
vacuum magnetic permeability $4\pi\alpha\hbar/e^2c$	μ_0	$1.256\,637\,061\,27(20) \times 10^{-6}$	N A^{-2}	1.6×10^{-10}
vacuum electric permittivity $1/\mu_0c^2$	ϵ_0	$8.854\,187\,8188(14) \times 10^{-12}$	F m^{-1}	1.6×10^{-10}
Josephson constant $2e/h$	K_{J}	$483\,597.848\,4 \dots \times 10^9$	Hz V^{-1}	exact
von Klitzing constant $\mu_0c/2\alpha = 2\pi\hbar/e^2$	R_{K}	25 812.807 45 ...	Ω	exact
magnetic flux quantum $2\pi\hbar/(2e)$	Φ_0	$2.067\,833\,848 \dots \times 10^{-15}$	Wb	exact
conductance quantum $2e^2/2\pi\hbar$	G_0	$7.748\,091\,729 \dots \times 10^{-5}$	S	exact
electron mass	m_{e}	$9.109\,383\,7139(28) \times 10^{-31}$	kg	3.1×10^{-10}
proton mass	m_{p}	$1.672\,621\,925\,95(52) \times 10^{-27}$	kg	3.1×10^{-10}
proton-electron mass ratio	$m_{\text{p}}/m_{\text{e}}$	1836.152 673 426(32)		1.7×10^{-11}
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297\,352\,5643(11) \times 10^{-3}$		1.6×10^{-10}
inverse fine-structure constant	α^{-1}	137.035 999 177(21)		1.6×10^{-10}
Rydberg frequency $\alpha^2m_{\text{e}}c^2/2\hbar$	cR_{∞}	$3.289\,841\,960\,2500(36) \times 10^{15}$	Hz	1.1×10^{-12}
Boltzmann constant	k	$1.380\,649 \times 10^{-23}$	J K^{-1}	exact
Avogadro constant	N_{A}	$6.022\,140\,76 \times 10^{23}$	mol^{-1}	exact
molar gas constant $N_{\text{A}}k$	R	8.314 462 618 ...	$\text{J mol}^{-1} \text{K}^{-1}$	exact
Faraday constant $N_{\text{A}}e$	F	96 485.332 12 ...	C mol^{-1}	exact
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670\,374\,419 \dots \times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$	exact
Non-SI units accepted for use with the SI				
electron volt (e/C) J	eV	$1.602\,176\,634 \times 10^{-19}$	J	exact
(unified) atomic mass unit $\frac{1}{12}m(^{12}\text{C})$	u	$1.660\,539\,068\,92(52) \times 10^{-27}$	kg	3.1×10^{-10}

* The energy of a photon with frequency ν expressed in unit Hz is $E = h\nu$ in J. Unitary time evolution of the state of this photon is given by $\exp(-iEt/\hbar)|\varphi\rangle$, where $|\varphi\rangle$ is the photon state at time $t = 0$ and time is expressed in unit s. The ratio $Et/\hbar$ is a phase.

1 eV
1.602 176 634 × 10⁻¹⁹ J

(1 eV) =
1.602 176 634 × 10⁻¹⁹ J

(1 eV)/c² =
1.782 661 921 ... × 10⁻³⁶ kg

(1 eV)/hc =
8.065 543 937 ... × 10⁵ m⁻¹

(1 eV)/h =
2.417 989 242 ... × 10¹⁴ Hz

1 eV
1.160 451 812 ... × 10⁴ K

1 eV ⇔ 11604.51812 K ⇔ 8065.543937 cm⁻¹ ⇔ 1239.841984 nm

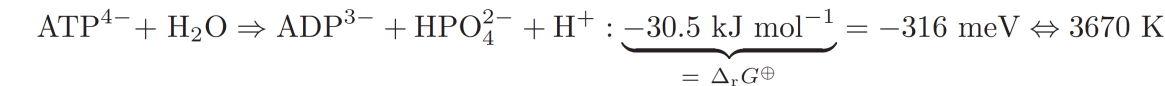
$\lambda = \frac{c}{\nu}, h\nu = \frac{hc}{\lambda}$

1 eV = 96.48533212 kJ mol⁻¹

1 kJ mol⁻¹ = 10.36426966 meV ⇔ 120.272 K

1 cal = 4.184 J, 1 kcal mol⁻¹ = 43.3641 meV ⇔ 503.218 K

300 K = 25.852 meV = 2.4943 kJ mol⁻¹



1 F = 1 C V⁻¹

$\frac{e^2}{4\pi\epsilon_0r} = \frac{[\text{C}^2]}{[\text{Fm}^{-1}\text{m}]} = [\text{CV}] = [\text{J}]$

1 D = c⁻¹ × 10⁻²¹ C m = 3.33564 × 10⁻³⁰ C m

Weber
 $\frac{\text{m}^2}{\text{A m}} = \text{T} = 10^4 \text{ Gauss}$

Dirac constant (exact)

$\hbar = \frac{h}{2\pi} = 6.582119569 \times 10^{-16} \text{ eV s}$

1 atm = 760 mmHg = 760 Torr = 1013.25 hPa = 1013.25 mbar

Fundamental Physical Constants — Atomic and nuclear constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
General				
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297\,352\,5643(11) \times 10^{-3}$		1.6×10^{-10}
inverse fine-structure constant	α^{-1}	137.035 999 177(21)		1.6×10^{-10}
Rydberg frequency $\alpha^2 m_e c^2/2h = E_h/2h$	cR_∞	$3.289\,841\,960\,2500(36) \times 10^{15}$	Hz	1.1×10^{-12}
energy equivalent	$hc\,R_\infty$	$2.179\,872\,361\,1030(24) \times 10^{-18}$	J	1.1×10^{-12}
		13.605 693 122 990(15)	eV	1.1×10^{-12}
Rydberg constant	R_∞	10 973 731.568 157(12)	[m ⁻¹]*	1.1×10^{-12}
Bohr radius $\hbar/\alpha m_e c = 4\pi\epsilon_0\hbar^2/m_e e^2$	a_0	$5.291\,772\,105\,44(82) \times 10^{-11}$	m	1.6×10^{-10}
Hartree energy $\alpha^2 m_e c^2 = e^2/4\pi\epsilon_0 a_0 = 2hcR_\infty$	E_h	$4.359\,744\,722\,2060(48) \times 10^{-18}$	J	1.1×10^{-12}
		27.211 386 245 981(30)	eV	1.1×10^{-12}
= 1 au = 2 Ryd				

2022年版(予定)	2023年予定	バージョン
[6]	定	x

新しいSIの定義に用いられる **プランク定数** h 、**電気素量** e 、**ボルツマン定数** k 、**アボガドロ定数** N_A の4つについて、最終的な数値の決定を**2018年1月28日**に出版した^[12]。これらの数値は「2017 special CODATA adjustment」と呼ばれている。

次に、新しいSIの定義の正式な採択に合わせて、新しい2018年版のCODATA推奨値となる「CODATA2018」を世界計量記念日の2019年5月20日に公表した。これは新しいSIの定義によって、様々な物理定数の数値（**不確かさ**の数値を含む）を変更する必要があったからである。