

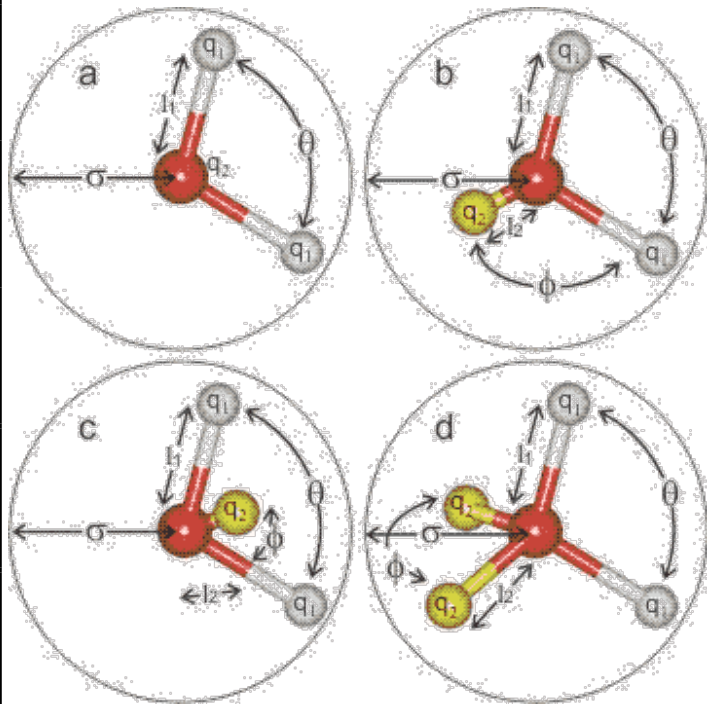
Liquid water MD simulation

A lot of stories on water:

Please refer to the excellent web-site
Martin Chaplin's "**Water Structure and Science**"

<http://www.lsbu.ac.uk/water/index2.html>

Model	Type	$\sigma / \text{\AA}$	$\epsilon / \text{kJ mol}^{-1}$	$l_1 \text{\AA}$	$l_2 \text{\AA}$	$q_1 (e)$	$q_2 (e)$	θ°	φ°
SPC/E	a	3.166	0.650	1.0000	-	+0.423 8	-0.847 6	109.47	-
TIP3P	a	3.1506 1	0.6364	0.9572	-	+0.417 0	-0.834 0	104.52	-
TIP4P	c	3.1536 5	0.6480	0.9572	0.15	+0.520 0	-1.040 0	104.52	52.26
TIP4P-Ew	c	3.1643 5	0.6809 46	0.9572	0.125	+0.524 22	-1.048 44	104.52	52.26
TIP5P-Ew	d	3.097	0.7448	0.9572	0.70	+0.241 0	-0.241 0	104.52	109.4 7



Water model potential

SPC/E: H. J. C. Berendsen, J. R. Grigera and T. P. Straatsma, The missing term in effective pair potentials, *J. Phys. Chem.* **91** (1987) 6269-6271.

TIP3P: W. L. Jorgensen, J. Chandrasekhar, J. D. Madura, R. W. Impey, and M. L. Klein, Comparison of simple potential functions for simulating liquid water, *J. Chem. Phys.* **79** (1983) 926-935

TIP4P: W. L. Jorgensen and J. D. Madura, Temperature and size dependence for monte carlo simulations of TIP4P water, *Mol. Phys.* **56** (1985) 1381-1392.

TIP4P-Ew: H. W. Horn, W. C. Swope, J. W. Pitera, J. D. Madura, T. J. Dick, G. L. Hura and T. Head-Gordon, Development of an improved four-site water model for biomolecular simulations: TIP4P-Ew, *J. Chem. Phys.* **120** (2004) 9665-9678.

TIP5P-Ew: S. W. Rick, A reoptimization of the five-site water potential (TIP5P) for use with Ewald sums, *J. Chem. Phys.* **120** (2004) 6085-6093.

TIP5P: M. W. Mahoney and W. L. Jorgensen, A five-site model for liquid water and the reproduction of the density anomaly by rigid, nonpolarizable potential functions, *J. Chem. Phys.* **112** (2000) 8910-8922.

Calculated results : classical MD

Model	Dipole moment	Dielectric constant	Self diffusion, $10^{-5} \text{ cm}^2/\text{s}$	Average configurational energy, kJ mol^{-1}	Density maximum, $^{\circ}\text{C}$	Expansion coefficient, $10^{-4} \text{ }^{\circ}\text{C}^{-1}$
Exp.	2.95	78.4	2.30	-41.5	+3.984	2.53
SPC/E	2.35	71	2.49	-41.5	-38	5.14
TIP3P	2.35	82	5.19	-41.1	-91	9.2
TIP4P	2.18	53	3.29	-41.8	-25	4.4
TIP4P-Ew	2.32	62.9	2.4	-46.5	+1	3.1
TIP5P-Ew	2.29	92	2.8	-	+8	4.9

All the data is at 25°C and 1 atm

==>The popular models SPC, SPC/E, TIP3P and TIP4P produce poor agreement with water's melting point (giving melting points of 190 K, 215 K, 146 K and 232 K respectively)

==>The popular TIP4P model underestimates the tetrahedrality of the water molecule's environment, which explains its poor estimate of the dielectric constant. It is, however, remarkably good at qualitatively describing water's phase diagram. The SPC/E and TIP4P models are reported as failing to properly describe the experimental $\text{O}\cdots\text{O}$ radial distribution function. The TIP3P and SPC show particularly poor agreement, the TIP4P, SPC/E and PPC show improved agreement.

OF COMPUTER SIMULATIONS ON WATER

Bertrand Guillot

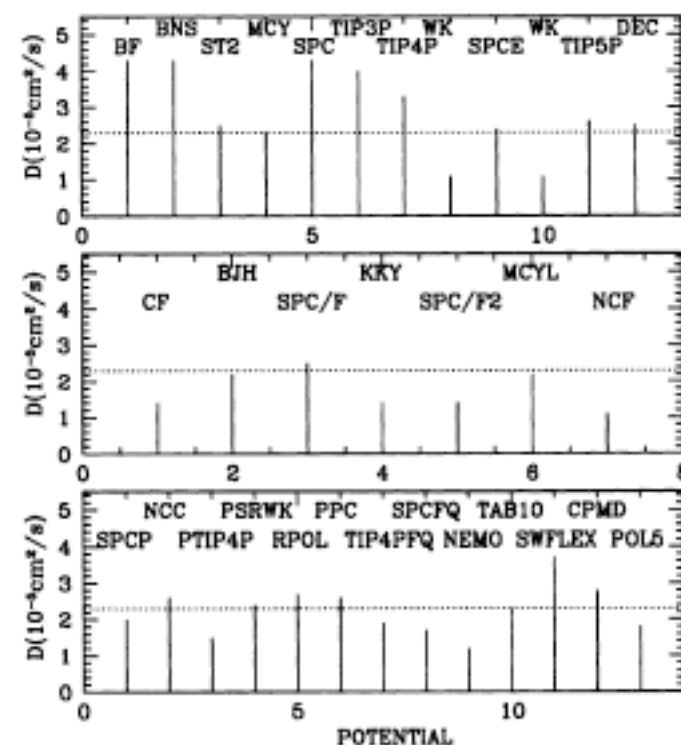
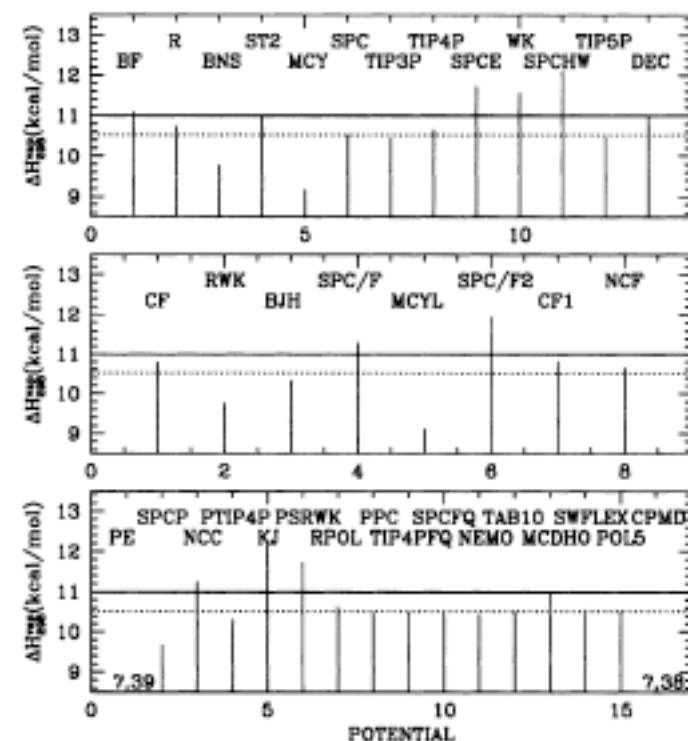
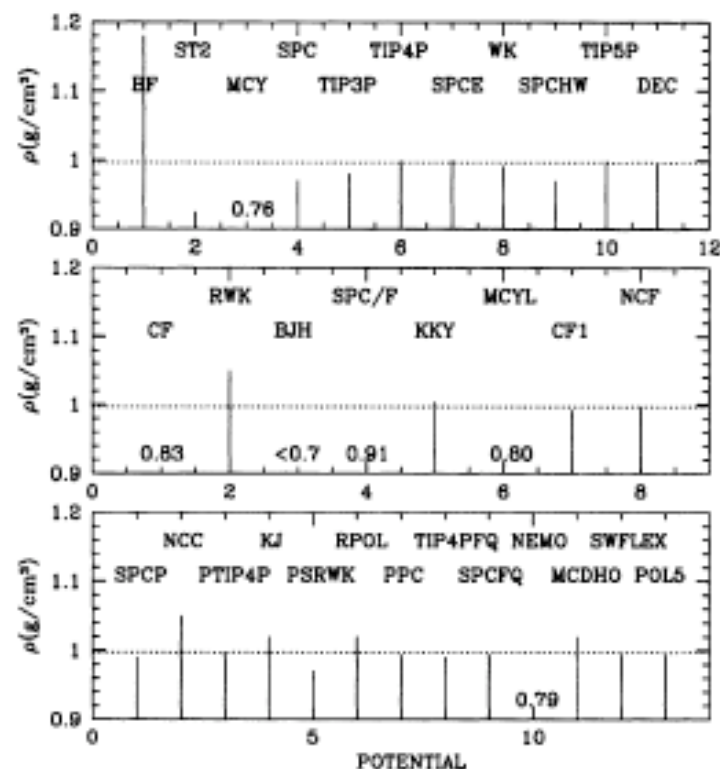
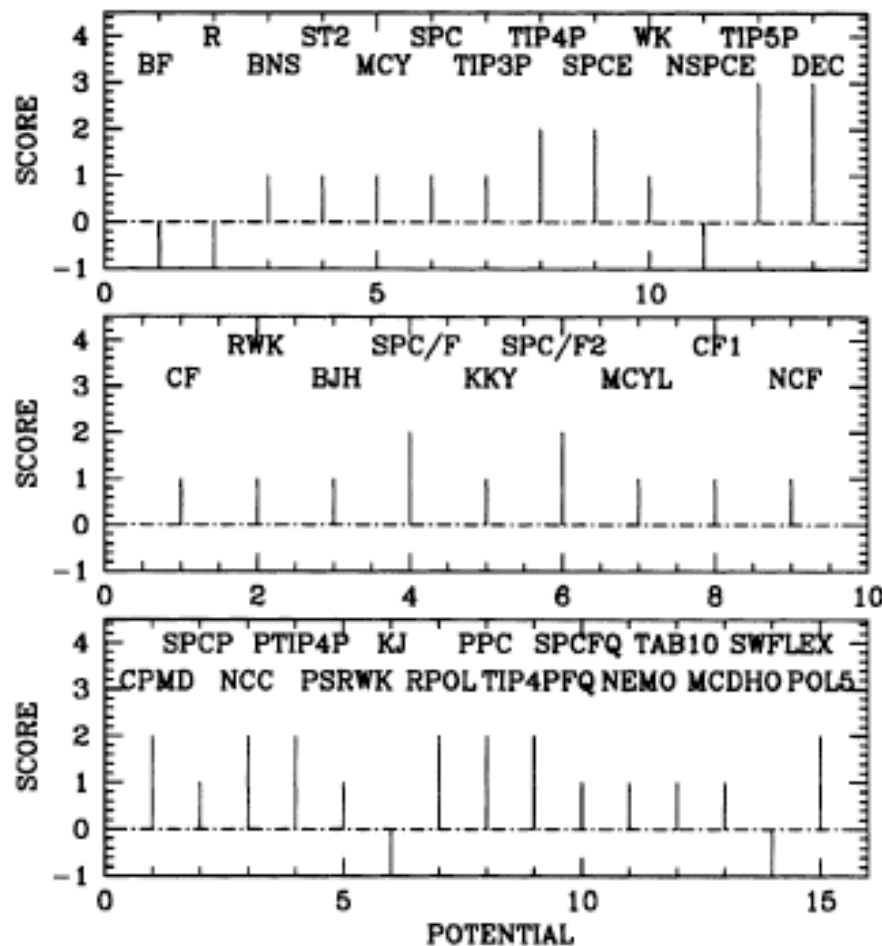


Figure 1. Density of simulated water at 298K and 1bar for different model potentials (see Table): top panel (rigid models), middle panel (flexible models) and bottom panel (polarizable models). The dotted line indicates the experimental value for H_2O (0.997 g/cm³).

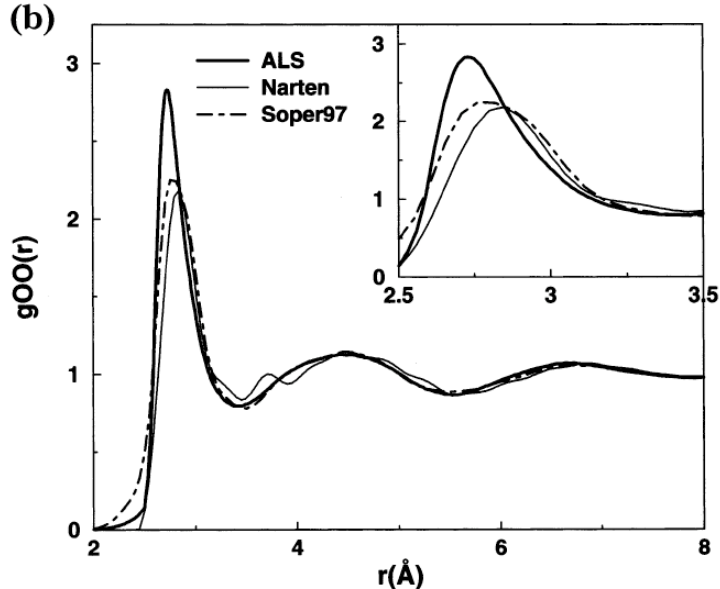
model potential SCORE



==> We will use
SPC/E
and
TIP4P/Ew
models.

Figure 5. Ability of the model potentials to reproduce structure data of liquid water at ambient conditions (see text): -1 suggests a bad description, +1 when it is acceptable, +2 when the description is good and +3 when the agreement is excellent (see Fig.4 for an illustration).

flexible model: time consuming and not improved from rigid model
Polarizable model: so far not significantly improved, but maybe important for ion solvation, for interface, and for ILs



line). (b) Comparison of experimentally derived $g_{OO}(r)$ data. ALS, X-ray⁴⁰ (thick solid line); Narten,⁷⁰ X-ray (thin solid line); Soper, Bruni, and Ricci,¹⁰³ neutron 1997 (dot-dash line).

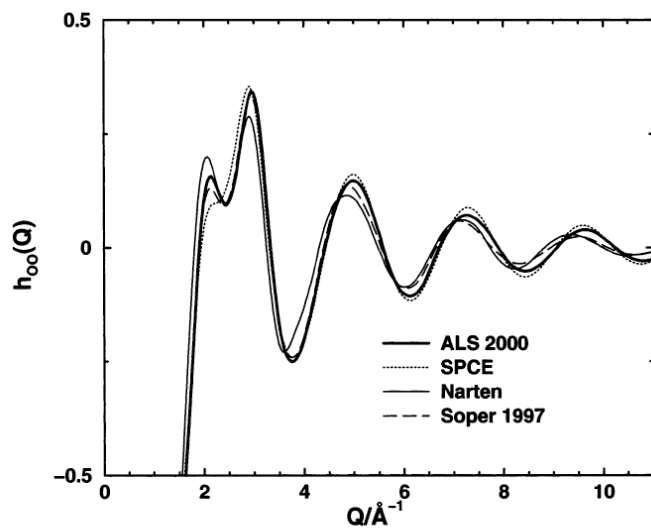


Figure 3. $h_{OO}(Q)$ from experiment and simulation. Legend: Narten and Levy⁶³ (thin solid line); Soper, Bruni, and Ricci¹⁰³ (dot-dash line); Hura et al.⁴⁰ (thick solid line); SPC/E²⁴ (dotted line). The curve for Narten and Levy is $H_M(Q)$ taken from their paper;⁶² the curve for Soper et al. is taken from applying a Fourier transform to the $g_{OO}(r)$ given in ref 103.

ALS2000 and Soper 2000

ALS:X-ray 2000
Narten: X-ray 1972
Soper: Neutron 97
Soper 2000:reanalysis of
1997 data

Chem. Rev. **2002**, *102*, 2651–2670

Water Structure from Scattering Experiments and Simulation

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good agreement !!
SPC/E and ALS2000

Radial distribution
function(RDF)
from exp.
and MD.

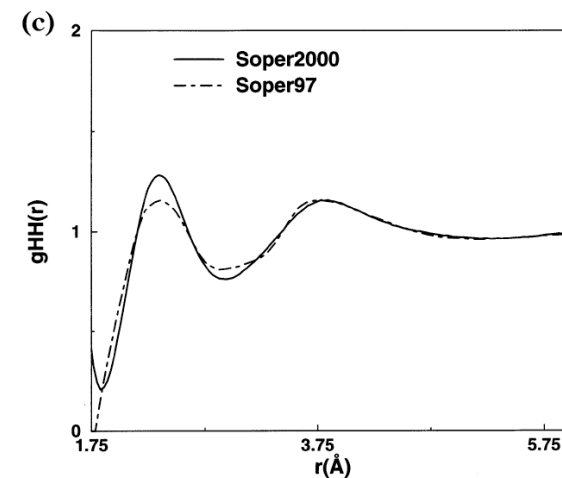
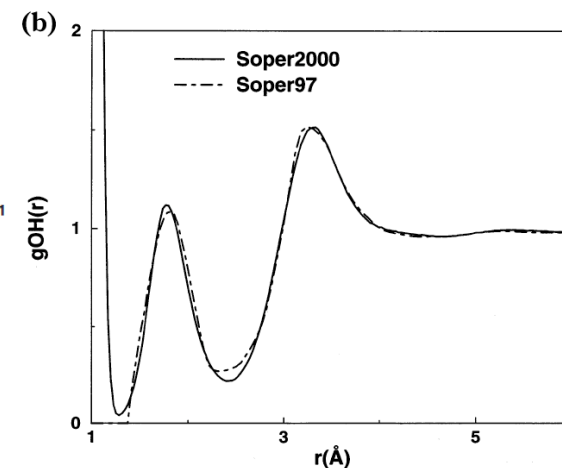
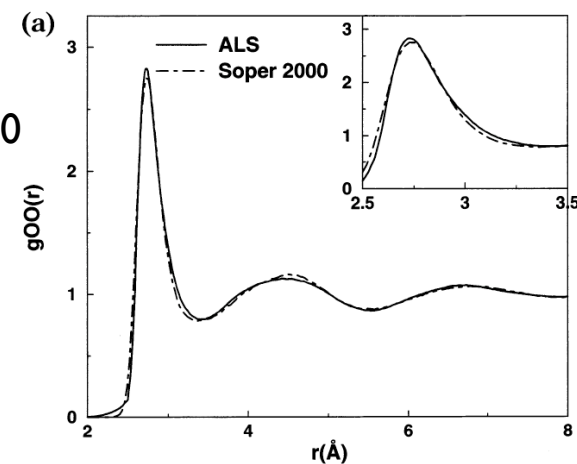


Figure 4. Comparison of neutron data on pure water at 25 °C and 1 atm. (a) Comparison of ALS X-ray experimental $g_{OO}(r)$ ^{40,41} (solid line) with reanalysis of Soper neutron data, 2000⁵ (dashed line). Comparison of neutron data on pure water at 25 °C and 1 atm in 1997¹⁰³ (dot-dash line) and 2000¹¹⁶ (solid line) for (b) $g_{OH}(r)$ and (c) $g_{HH}(r)$.

Classical model potentials can explain the Exp. results very well.

Exp. results are also improving.

Radial distribution function from exp. and MD.

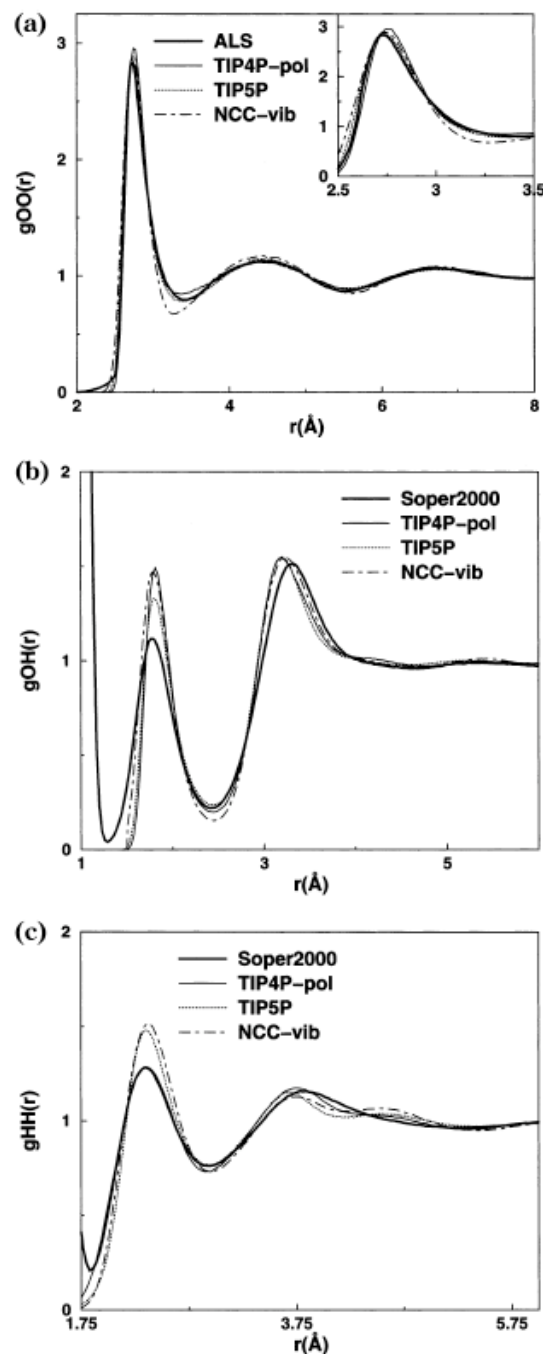


Figure 5. Comparison of experimental radial distribution functions (thick solid line) with simulations using empirical force fields for the TIP4P-pol¹³¹ (thin solid line), TIP5P²⁷ (dotted line), and NCC-vib²⁹ (dot-dash line) models: (a) $g_{OO}(r)$, (b) $g_{OH}(r)$, and (c) $g_{HH}(r)$.

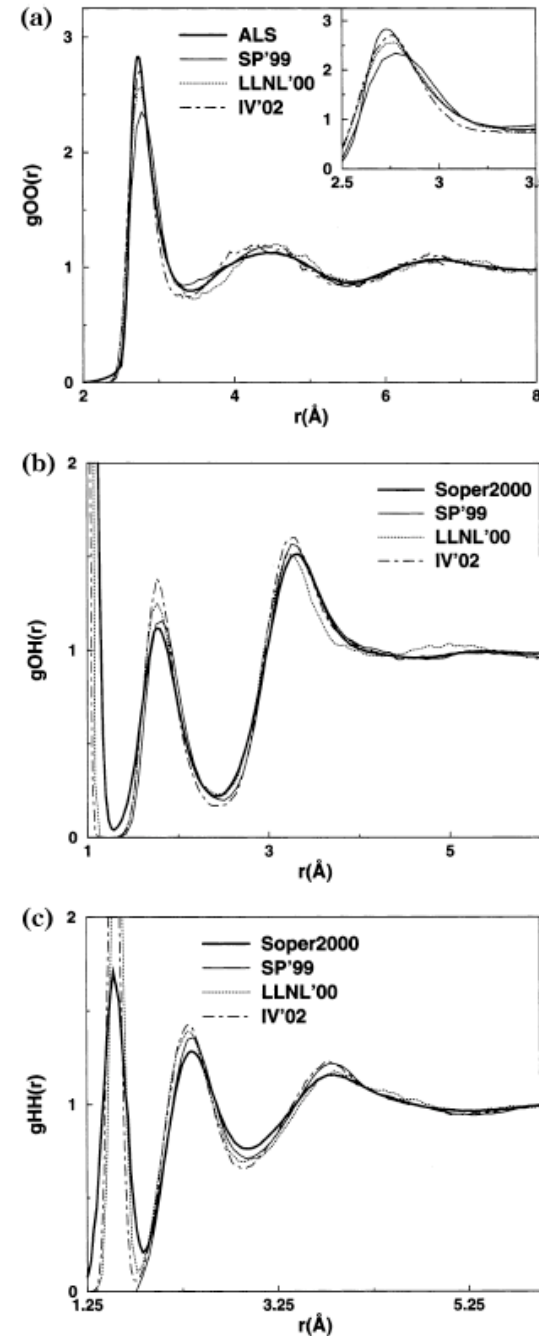


Figure 6. Comparison of experimental scattering data with ab initio molecular dynamics simulations. ALS X-ray experimental $g_{OO}(r)$ (thick solid line), ab initio simulation of 10 ps for 64 water molecules, average ionic temperature of 318 K from Silvestrelli and Parrinello³⁶ (thin solid line), recent ab initio simulation by Schwegler et al.¹⁸⁷ 3 ps for 54 water molecule average ionic temperature of ~ 294 K (dotted line), ab initio simulation of 12 ps for 64 water molecules, average ionic temperature of 307 K from Izvekov and Voth³⁸ (dot-dash line): (a) $g_{OO}(r)$, (b) $g_{OH}(r)$, (c) $g_{HH}(r)$.

first-principles MD is getting better.

Here we use one of the water model SPC/E or TIP4P-Ew

1) L-J interaction is oxygen-oxygen only.

2) rigid molecule is assumed. $\Rightarrow$ Cal. become faster

O-H distance is const.

angle H-O-H is const.

$\Rightarrow$ Cal. become faster

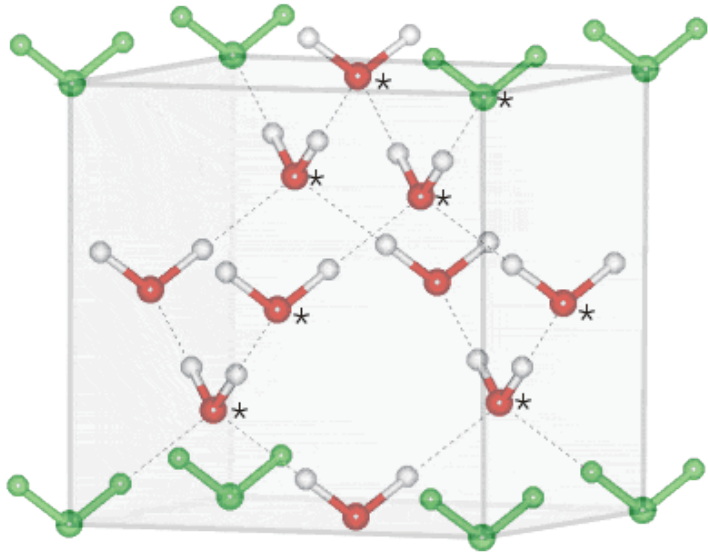
$\Rightarrow$ but neglect the vibrational degree of freedom.

$\Rightarrow$ we should select how we impose the constraints on bond length and bond angle.

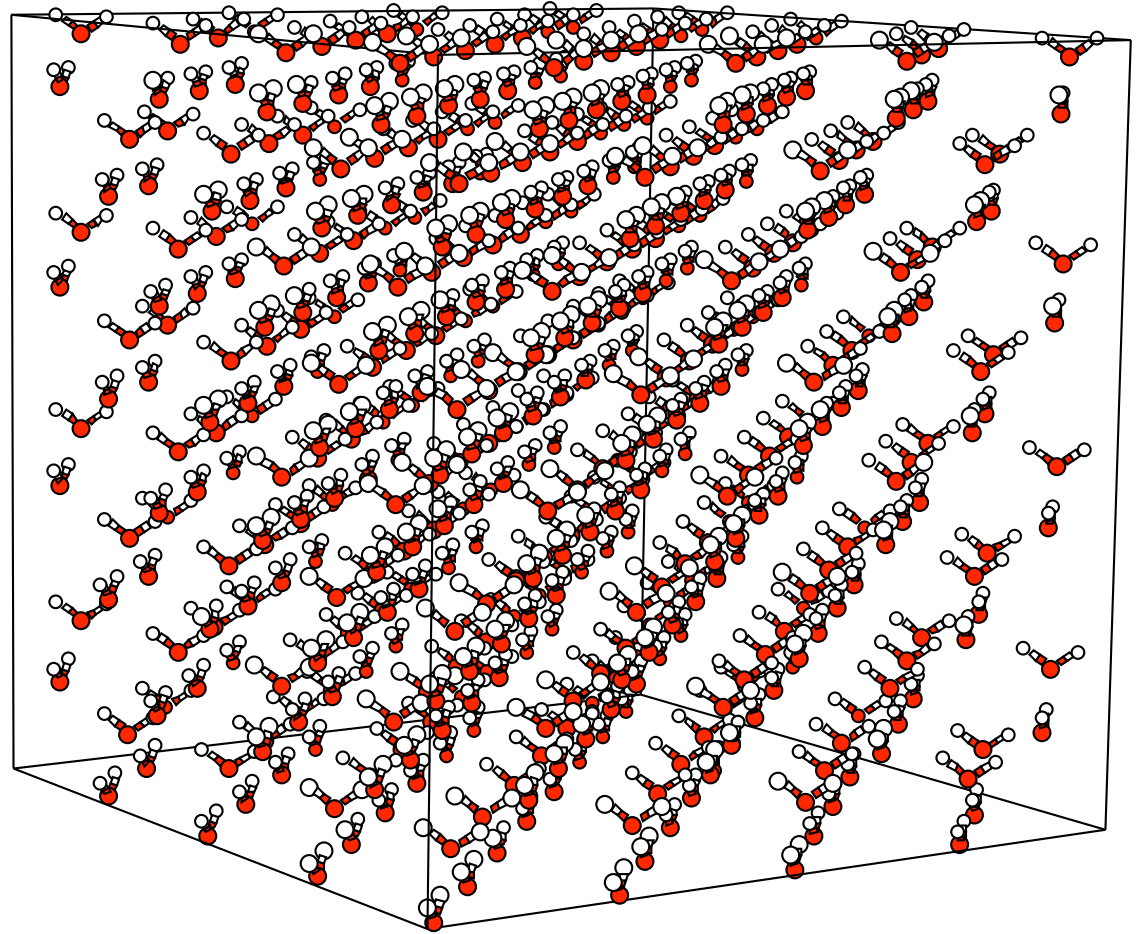
3) no electrostatic interaction between the intramolecular atoms.

initial configuration: ice?

Cubic Ice



fcc + half the tetrahedral sites are filled.



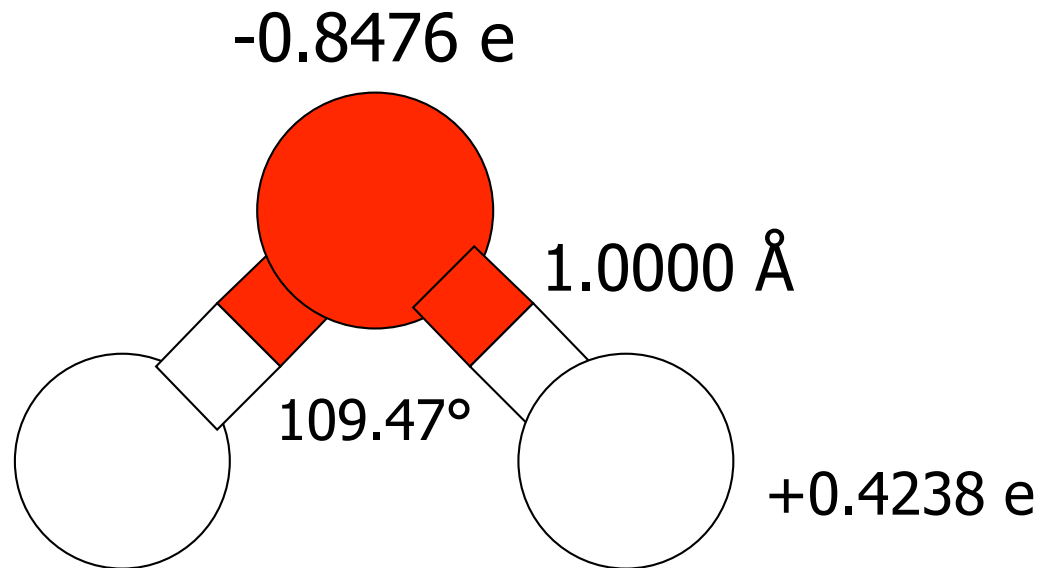
$$\rho / \text{g cm}^{-3} = 0.996516$$

H2O all site(fcc+t-site)= 512

liquid water all atom number= 1536

ax,ay,az,volume= 24.86, 24.86, 24.86 Å 15370.0

SPC/E model



$r(\text{OH})=1.0000\ \text{\AA}$
 $\theta(\text{HOH})=109.47$
are fixed.

intermolecular interaction

$z(\text{O})=-0.8476\ e$
 $z(\text{H})=0.4238\ e$
 $\epsilon(\text{O}) = 0.650\ \text{kJ mol}^{-1}$
 $\sigma(\text{O}) = 3.166\ \text{\AA}$

CONFIG

H2O liquid SPCE water at 300 K

2	1	
24.8632886676	.0000000000	.0000000000
.0000000000	24.8632886676	.0000000000
.0000000000	.0000000000	24.8632886676
OW 1		
-12.4316443338	-12.4316443338	-12.4316443338
-.1157118156	-.1006883860	.0252069353
.0000000000	.0000000000	.0000000000
HW 2		
-11.8542984133	-11.8542984133	-11.8542853673
-.1157118156	-.1006883860	.0252069353
.0000000000	.0000000000	.0000000000
HW 3		
-13.0089902543	-13.0089902543	-11.8542853673
-.1157118156	-.1006883860	.0252069353
.0000000000	.0000000000	.0000000000
OW 4		
-10.8776887921	-10.8776887921	-10.8776887921
-.0944489040	.0834090526	-.0070384049
.0000000000	.0000000000	.0000000000
HW 5		
-10.3003428716	-11.4550347126	-10.3003298256
-.0944489040	.0834090526	-.0070384049
.0000000000	.0000000000	.0000000000
HW 6		
-11.4550347126	-10.3003428716	-10.3003298256
-.0944489040	.0834090526	-.0070384049
.0000000000	.0000000000	.0000000000
...		
OW 1534		
7.7697777086	10.8776887921	10.8776887921
.0179460404	.0359122702	.0141856681
.0000000000	.0000000000	.0000000000
HW 1535		
8.3471236291	10.3003428716	11.4550477586
.0179460404	.0359122702	.0141856681
.0000000000	.0000000000	.0000000000
HW 1536		
7.1924317881	11.4550347126	11.4550477586
.0179460404	.0359122702	.0141856681
.0000000000	.0000000000	.0000000000

FIELD

H2O SPC/E		
units kJ		
molecules 1		
SPCE water		
nummols 512		
atoms 3		
OW	15.9996	-0.8476
HW	1.0008	0.4238
HW	1.0008	0.4238
constraints 3		
1 2	1.0000	
1 3	1.0000	
2 3	1.63298	
finish		
vdW 3		
OW	OW lj	0.6507346 3.165492
OW	HW lj	0.0 3.165492
HW	HW lj	0.0 0.0
close		

SPC/E model

CONTROL NVE

SPC/E model of water

integrator velocity verlet
temperature 300.00
ensemble nve

steps 150000
equilibration 100000
multiple step 1
scale 10
print 50
stack 100
stats 10
rdf 1
traj 1 100 0

timestep 0.001
cutoff 9.50
delr width 0.5000
ewald precision 1d-6
shake tolerance 1.0E-5
quaternion tolerance 1.0E-5
print rdf

job time 150000.00
close time 500.00

finish

CONTROL NPT 300 K 1 atom

SPC/E model of water

integrator velocity verlet
temperature 300.00
ensemble npt hoover 0.1 0.2

steps 150000
equilibration 100000
multiple step 1
scale 10
print 50
stack 100
stats 10
rdf 1
traj 1 100 0

timestep 0.001
cutoff 9.50
delr width 0.5000
ewald precision 1d-6
shake tolerance 1.0E-5
quaternion tolerance 1.0E-5
print rdf
pres 0.001

job time 150000.00
close time 500.00

finish

SPC/E model

NVE : microcanonical ensemble

$\Rightarrow$ total energy conserved

$\Rightarrow$ directly linked to mechanics

$\Rightarrow k_B T = \text{kinetic energy}$

Temperature fluctuates

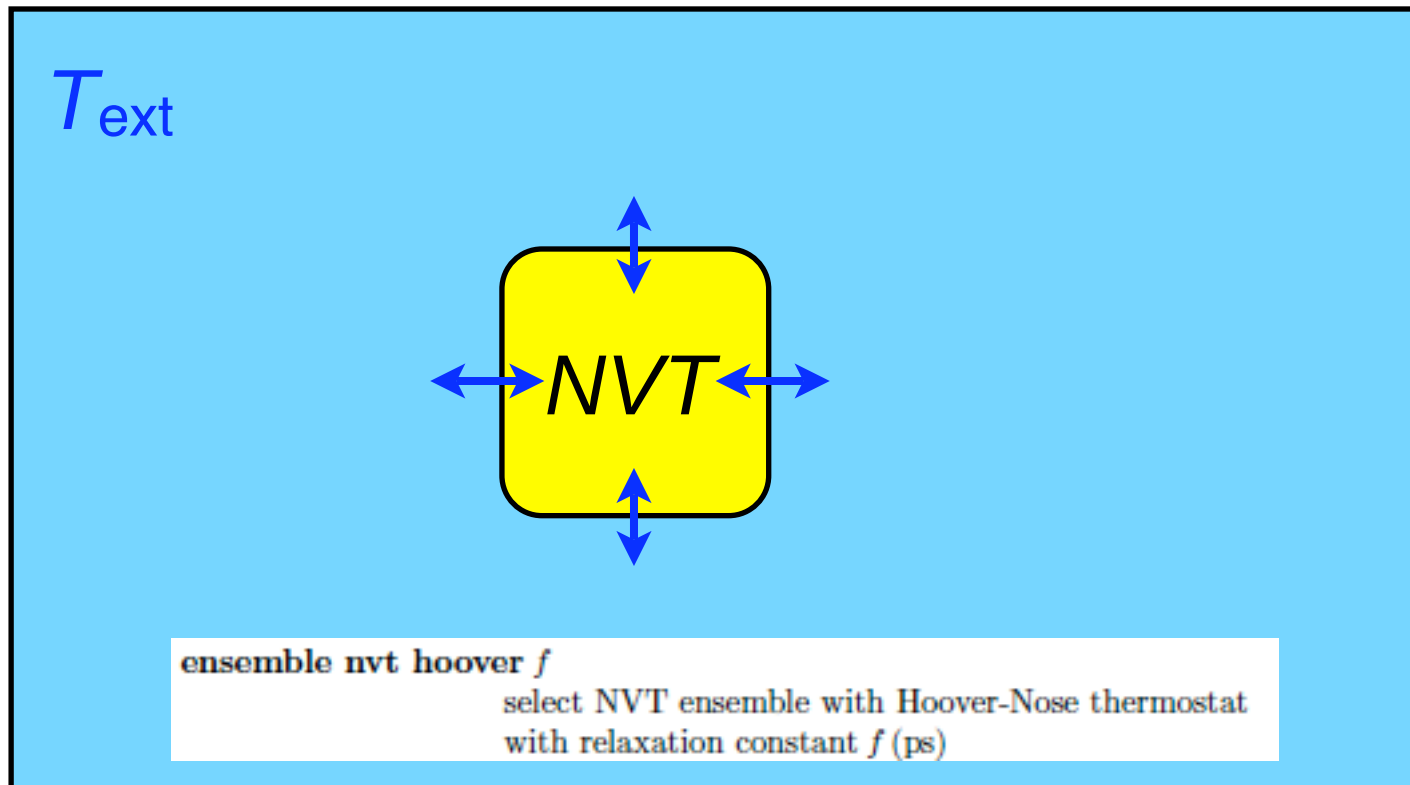
but the thermal average should be the same

NVT : canonical ensemble

$\Rightarrow T$ keeps constant and
the heat exchange (or momentum exchange)
with the virtual reservoir (thermostat)

$\Rightarrow$ some tricks are needed.

Nose-Hoover algorithm is only well defined!!



ensemble nvt hoover *f*

select NVT ensemble with Hoover-Nose thermostat
with relaxation constant *f* (ps)

2.5.4.1 Nosé - Hoover Thermostat

In the Nosé-Hoover algorithm [20] Newton's equations of motion are modified to read:

$$\begin{aligned}\frac{d\mathbf{r}(t)}{dt} &= \mathbf{v}(t) \\ \frac{d\mathbf{v}(t)}{dt} &= \frac{\mathbf{f}(t)}{m} - \chi(t)\mathbf{v}(t)\end{aligned}\tag{2.231}$$

$$\tag{2.232}$$

The friction coefficient, χ , is controlled by the first order differential equation

$$\frac{d\chi(t)}{dt} = \frac{N_f k_B}{Q} (T(t) - T_{\text{ext}}) \tag{2.233}$$

where $Q = N_f k_B T_{\text{ext}} \tau_T^2$ is the effective 'mass' of the thermostat, τ_T is a specified time constant (normally in the range [0.5, 2] ps) and N_f is the number of degrees of freedom in the system. $T(t)$ is the instantaneous temperature of the system at time t .

In the VV version of DL_POLY_2

$$\chi(t + \frac{1}{2}\Delta t) \leftarrow \chi(t) + \frac{\Delta t N_f k_B}{2Q} (T(t) - T_{\text{ext}})$$

$$\mathbf{v}'(t) \leftarrow \mathbf{v}(t) - \frac{\Delta t}{2} \chi(t + \frac{1}{2}\Delta t) \mathbf{v}(t)$$

$$\mathbf{v}(t + \frac{1}{2}\Delta t) \leftarrow \mathbf{v}'(t) + \frac{\Delta t}{2} \frac{\mathbf{f}(t)}{m}$$

$$\mathbf{r}(t + \Delta t) \leftarrow \mathbf{r}(t) + \Delta t \mathbf{v}(t + \frac{1}{2}\Delta t)$$

call rattle(R)

$$\mathbf{v}'(t + \Delta t) \leftarrow \mathbf{v}(t + \frac{1}{2}\Delta t) + \frac{\Delta t}{2} \frac{\mathbf{f}(t + \Delta t)}{m}$$

call rattle(V)

$$\chi(t + \Delta t) \leftarrow \chi(t + \frac{1}{2}\Delta t) + \frac{\Delta t N_f k_B}{2Q} (T(t + \Delta t) - T_{\text{ext}})$$

$$\mathbf{v}(t + \Delta t) \leftarrow \mathbf{v}'(t + \Delta t) - \frac{\Delta t}{2} \chi(t + \Delta t) \mathbf{v}'(t + \Delta t)$$

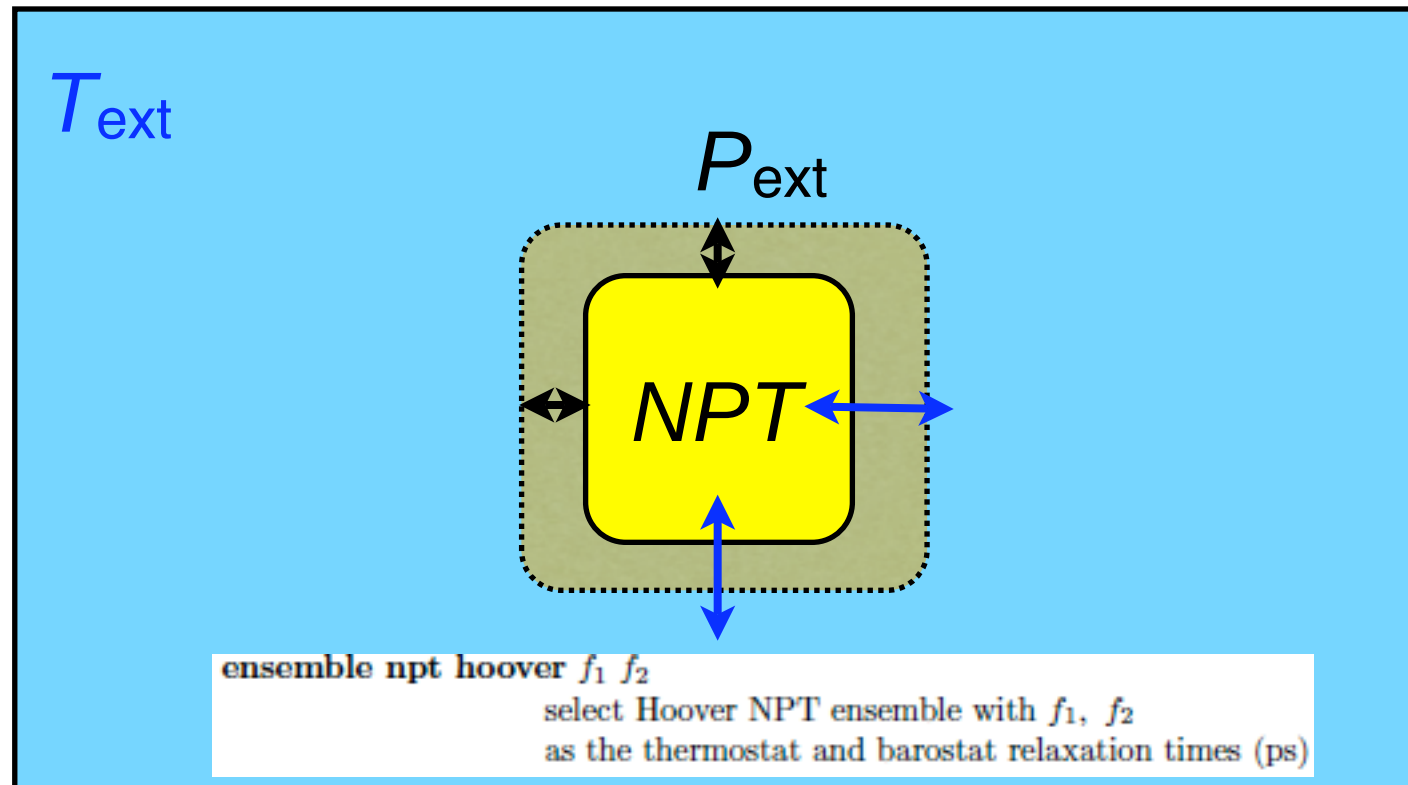
The conserved quantity is derived from the extended Hamiltonian for the system which, to within a constant, is the Helmholtz free energy:

$$\mathcal{H}_{\text{NVT}} = U + KE + \frac{1}{2} Q \chi(t)^2 + \frac{Q}{\tau_T^2} \int_0^t \chi(s) ds$$

NPT ensemble

$\Rightarrow T$ and P keeps constant and
the heat exchange and cell size change
dynamically.

$\Rightarrow$ Hoover barostat method is only well defined!!



ensemble npt hoover f_1 f_2
 select Hoover NPT ensemble with f_1 , f_2
 as the thermostat and barostat relaxation times (ps)

Cell size variation

For isotropic fluctuations the equations of motion are:

$$\begin{aligned}
 \frac{d\mathbf{r}(t)}{dt} &= \mathbf{v}(t) + \eta(\mathbf{r}(t) - \mathbf{R}_0) \\
 \frac{d\mathbf{v}(t)}{dt} &= \frac{\mathbf{f}(t)}{m} - [\chi(t) + \eta(t)] \mathbf{v}(t) \\
 \frac{d\chi(t)}{dt} &= \frac{N_f k_B}{Q} (T(t) - T_{\text{ext}}) + \frac{1}{Q} (W \eta(t)^2 - k_B T_{\text{ext}}) \\
 \frac{d\eta(t)}{dt} &= \frac{3}{W} V(t) (\mathcal{P}(t) - P_{\text{ext}}) - \chi(t) \eta(t) \\
 \frac{dV(t)}{dt} &= [3\eta(t)] V(t)
 \end{aligned} \tag{2.246}$$

where $Q = N_f k_B T_{\text{ext}} \tau_T^2$ is the effective ‘mass’ of the thermostat and $W = N_f k_B T_{\text{ext}} \tau_P^2$ is the effective ‘mass’ of the barostat. N_f is the number of degrees of freedom, η is the barostat friction coefficient, $\mathbf{R}_0$ the system centre of mass, τ_T and τ_P are specified time constants for temperature and pressure fluctuations respectively, $\mathcal{P}(t)$ is the instantaneous pressure and V the system volume.

The conserved quantity is, to within a constant, the Gibbs free energy of the system:

$$\mathcal{H}_{NPT} = U + KE + \mathcal{P}_{\text{ext}} V(t) + \frac{1}{2} Q \chi(t)^2 + \frac{1}{2} W \eta(t)^2 + \int_0^t \left(\frac{Q}{\tau_T^2} \chi(s) + k_B T_{\text{ext}} \right) ds \tag{2.247}$$

The implementation in the VV algorithm follows the scheme:

$$\begin{aligned}
\chi(t + \frac{1}{2}\Delta t) &\leftarrow \chi(t) + \frac{\Delta t N_f k_B}{2Q} (\mathcal{T}(t) - T_{\text{ext}}) + \frac{\Delta t}{2Q} (W \eta(t)^2 - k_B T_{\text{ext}}) \\
\underline{v}'(t) &\leftarrow \underline{v}(t) - \frac{\Delta t}{2} \chi(t + \frac{1}{2}\Delta t) \underline{v}(t) \\
\eta(t + \frac{1}{2}\Delta t) &\leftarrow \eta(t) + \frac{\Delta t}{2} \left\{ \frac{3V(t)}{W} (\mathcal{P}(t) - P_{\text{ext}}) - \chi(t) \eta(t) \right\} \\
\underline{v}''(t) &\leftarrow \underline{v}'(t) - \frac{\Delta t}{2} \eta(t + \frac{1}{2}\Delta t) \underline{v}'(t) \\
\underline{v}(t + \frac{1}{2}\Delta t) &\leftarrow \underline{v}''(t) + \frac{\Delta t}{2} \frac{\underline{f}(t)}{m} \\
\underline{r}(t + \Delta t) &\leftarrow \underline{r}(t) + \Delta t \underline{v}(t + \frac{1}{2}\Delta t) \\
&\text{call } \text{rattle}(R) \\
V(t + \Delta t) &\leftarrow V(t) \exp \left[3\Delta t \eta(t + \frac{1}{2}\Delta t) \right] \\
\underline{\underline{\mathbf{H}}}(t + \Delta t) &\leftarrow \exp \left[\Delta t \eta(t + \frac{1}{2}\Delta t) \right] \underline{\underline{\mathbf{H}}}(t) \\
\underline{v}'(t + \Delta t) &\leftarrow \underline{v}(t + \frac{1}{2}\Delta t) + \frac{\Delta t}{2} \frac{\underline{f}(t + \Delta t)}{m} \\
&\text{call } \text{rattle}(V) \\
\eta(t + \Delta t) &\leftarrow \eta(t + \frac{1}{2}\Delta t) + \frac{\Delta t}{2} \left\{ \frac{V(t + \Delta t)}{W} (\mathcal{P}(t + \Delta t) - P_{\text{ext}}) - \chi(t + \Delta t) \eta(t + \Delta t) \right\} \\
\underline{v}''(t + \Delta t) &\leftarrow \underline{v}'(t + \Delta t) - \frac{\Delta t}{2} \eta(t + \Delta t) \underline{v}'(t + \Delta t) \\
\chi(t + \Delta t) &\leftarrow \chi(t + \frac{1}{2}\Delta t) + \frac{\Delta t N_f k_B}{2Q} (\mathcal{T}(t + \Delta t) - T_{\text{ext}}) + \frac{\Delta t}{2Q} (W \eta(t + \Delta t)^2 - k_B T_{\text{ext}}) \\
\underline{v}(t + \Delta t) &\leftarrow \underline{v}''(t + \Delta t) + \frac{\Delta t}{2} \chi(t + \Delta t) \underline{v}''(t + \Delta t) \tag{2.250}
\end{aligned}$$

SPC/E model: NVE ensemble results

time elapsed since job start = 27144.723 seconds

temp 2.9618E+02 K r.m.s. fluctuation 6.3217E+00 K

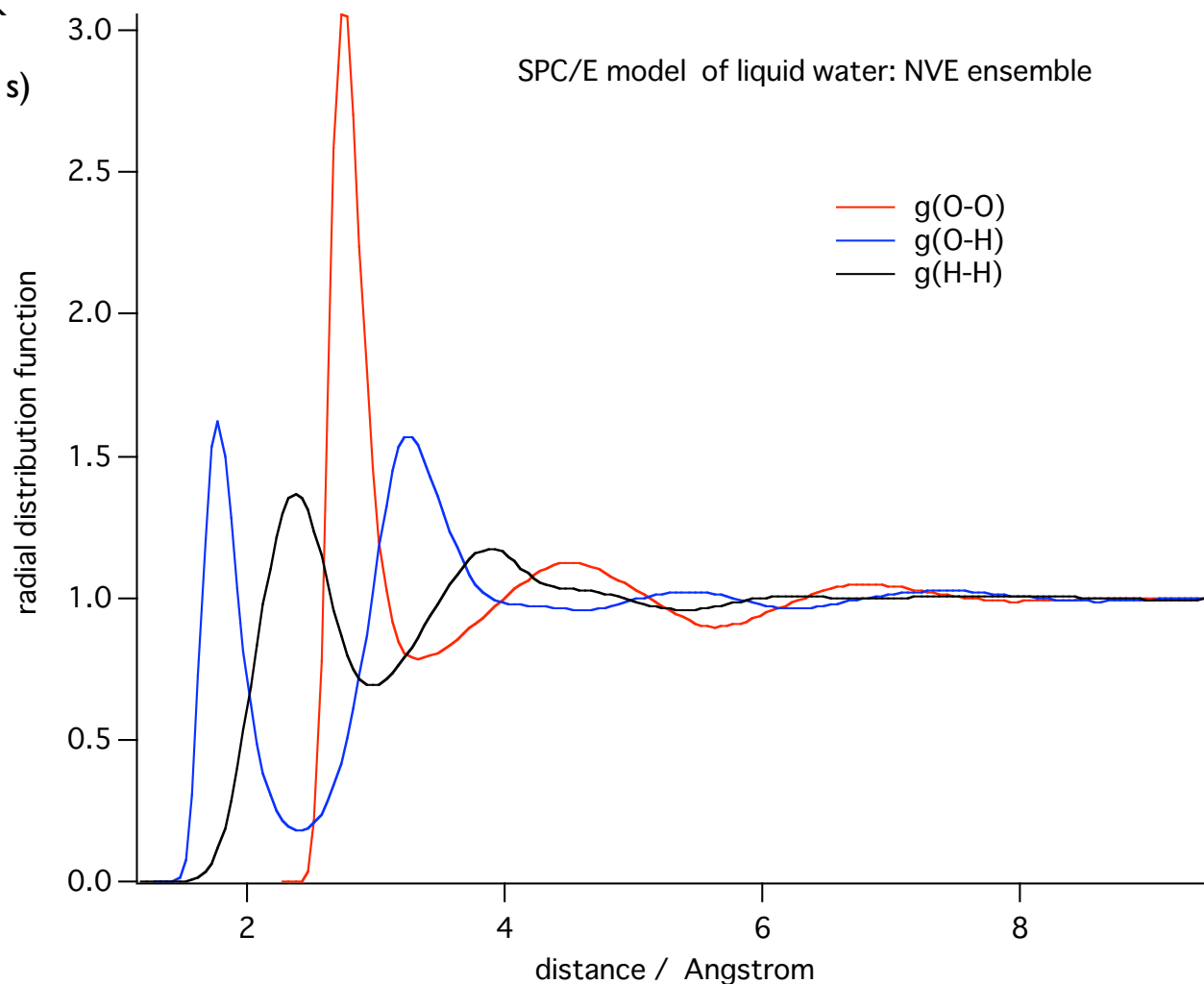
Approximate 3D Diffusion coefficients ($10^{-9} \text{ m}^2 / \text{s}$)

atom	D
OW	2.6341E+00
HW	2.6341E+00

Average pressure tensor

-8.2421E-02	2.9661E-02	5.1028E-03
2.9661E-02	-3.2231E-02	2.5959E-02
5.1028E-03	2.5959E-02	-8.3213E-02

trace/3. -6.5955E-02 katom

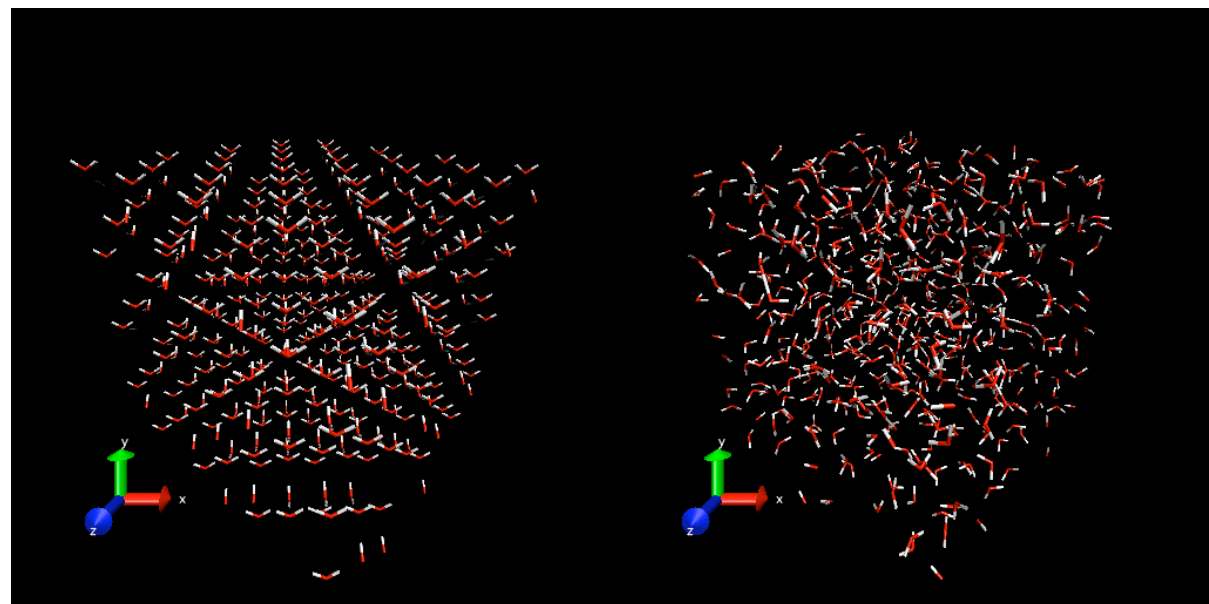


SPC/E model: NPT ensemble results

T = 3.0011E+02 K r.m.s. fluc. 7.1254E+00 K

Approximate 3D Diffusion coefficients (10^{-9} m² / s)

atom	D
OW	2.4579E+00
HW	2.4818E+00



movie

Average pressure 1.2925E-03 katom

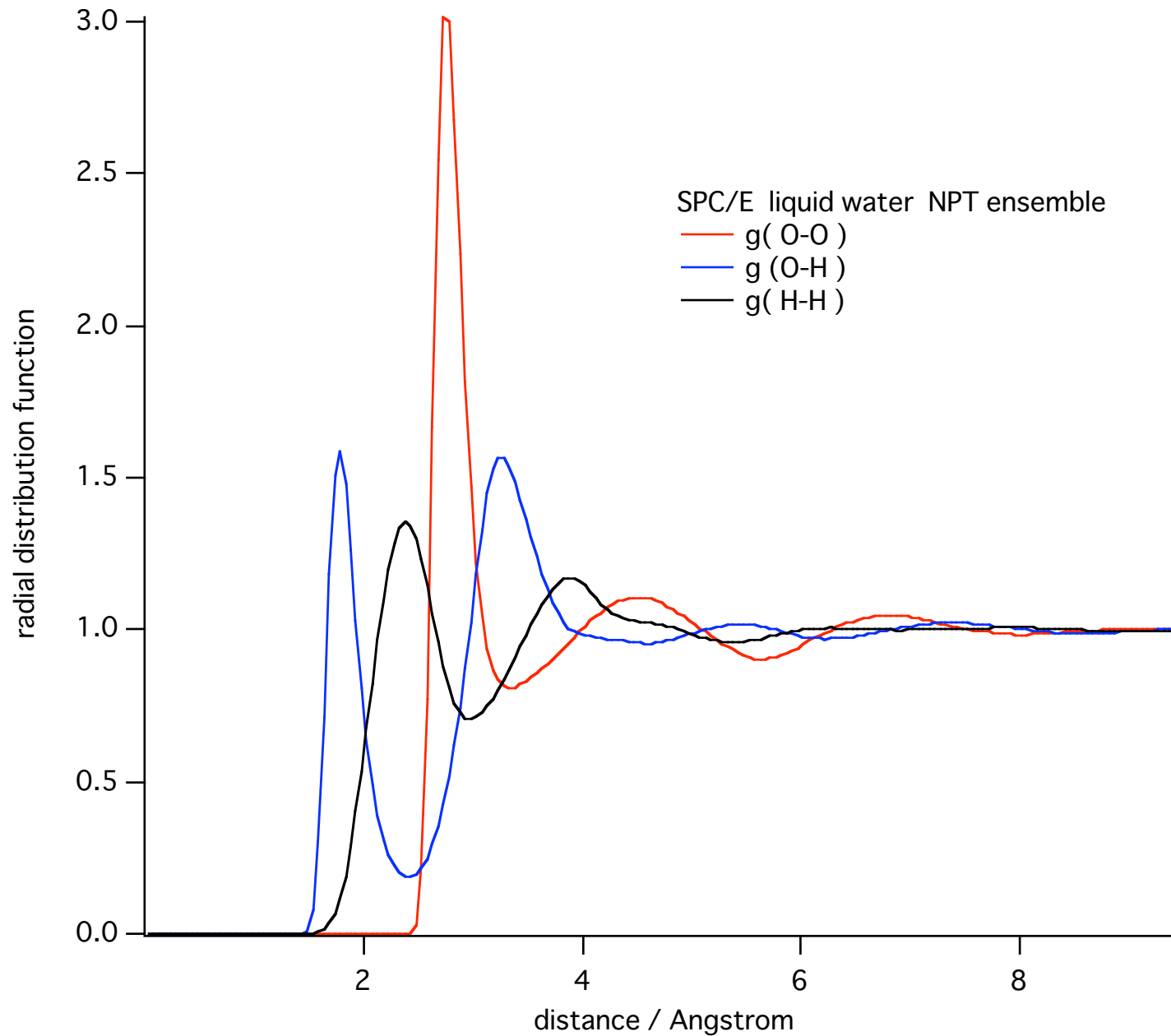
Average pressure tensor			r.m.s. fluctuations		
-3.3618E-02	1.3665E-02	3.6790E-02	6.2210E-01	5.0643E-01	5.1488E-01
1.3665E-02	-1.0684E-02	5.6640E-03	5.0643E-01	6.1777E-01	5.0854E-01
3.6790E-02	5.6640E-03	4.8179E-02	5.1488E-01	5.0854E-01	6.2829E-01
trace/3. 1.2925E-03					

Average cell vectors			r.m.s. fluctuations		
24.8372539181	.0000000000	.0000000000	7.1583E-02	0.0000E+00	0.0000E+00
.0000000000	24.8372539181	.0000000000	0.0000E+00	7.1583E-02	0.0000E+00
.0000000000	.0000000000	24.8372539181	0.0000E+00	0.0000E+00	7.1583E-02

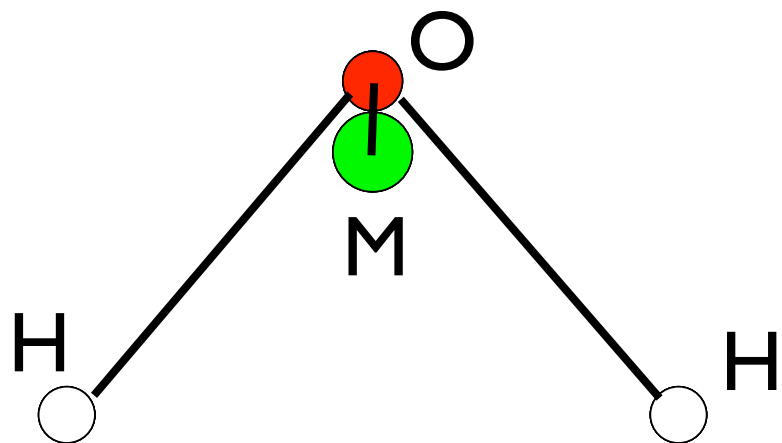
time elapsed since job start = 24121.500 seconds

SPC/E model:
NPT ensemble
results

cf. exp. results



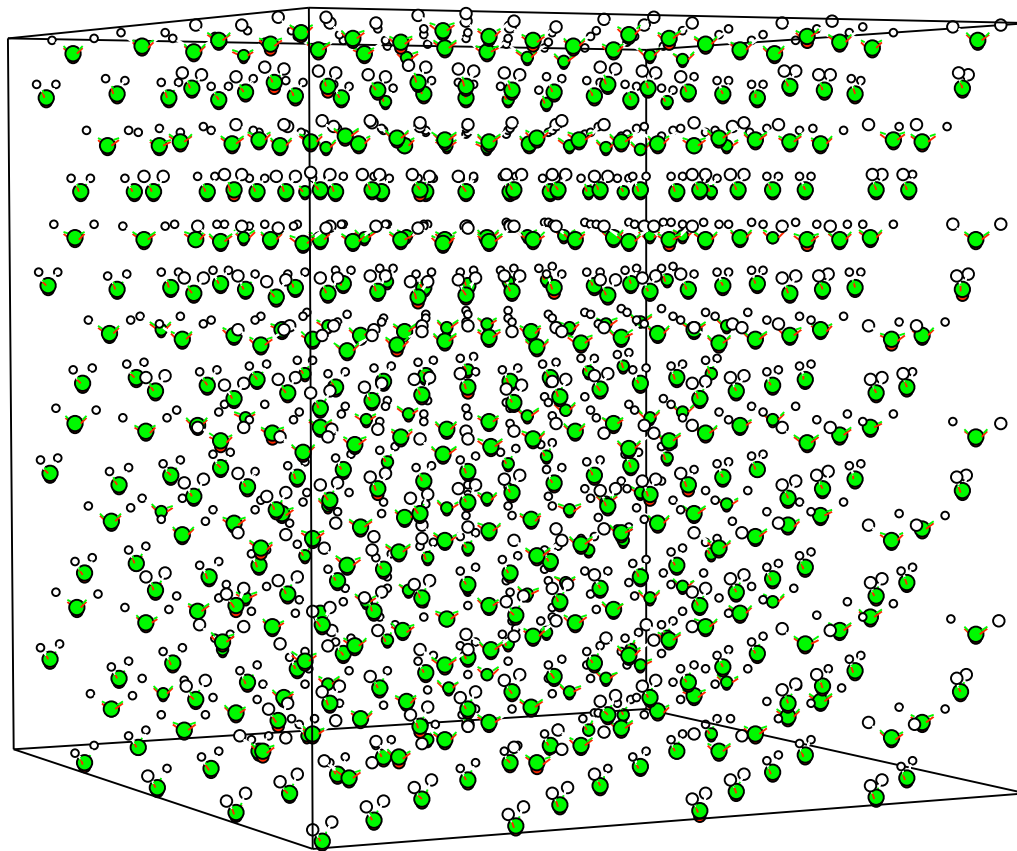
TIP4P/Ew model



M: dummy charge site

$r(\text{OH})=0.9572 \text{ \AA}$
 $r(\text{OM})=0.125 \text{ \AA}$
 $\theta(\text{HOH})=104.52$
 $\theta(\text{MOH})=52.26$
are fixed.

$z(\text{M})=-1.04844 e$
 $z(\text{H})=0.52422 e$
 $\epsilon(\text{O}) = 0.680946 \text{ kJ mol}^{-1}$
 $\sigma(\text{O}) = 3.16435 \text{ \AA}$
mass of M is zero.



CONFIG

TIP4P H2O liquid water 300 K

2	1	
24.8632889271	.0000000000	.0000000000
.0000000000	24.8632889271	.0000000000
.0000000000	.0000000000	24.8632889271
OW 1		
-12.4316444636	-12.4316444636	-12.4316444636
-.1157118180	-.1006883881	.0252069358
.0000000000	.0000000000	.0000000000
HW 2		
-11.8963997541	-11.8963997541	-11.8457621869
-.1157118180	-.1006883881	.0252069358
.0000000000	.0000000000	.0000000000
HW 3		
-12.9668891730	-12.9668891730	-11.8457621869
-.1157118180	-.1006883881	.0252069358
.0000000000	.0000000000	.0000000000
MW 4		
-12.4316444636	-12.4316444636	-12.3066444636
-.1157118180	-.1006883881	.0252069358
.0000000000	.0000000000	.0000000000
OW 5		
-10.8776889056	-10.8776889056	-10.8776889056
-.0944489059	.0834090543	-.0070384050
.0000000000	.0000000000	.0000000000
HW 6		
-10.3424441962	-11.4129336150	-10.2918066290
-.0944489059	.0834090543	-.0070384050
.0000000000	.0000000000	.0000000000
HW 7		
-11.4129336150	-10.3424441962	-10.2918066290
-.0944489059	.0834090543	-.0070384050
.0000000000	.0000000000	.0000000000
MW 8		
-10.8776889056	-10.8776889056	-10.7526889056
-.0944489059	.0834090543	-.0070384050
.0000000000	.0000000000	.0000000000

FIELD

H2O TIP4P				
units kj				
molecules 1				
TIP4P water				
nummols 512				
atoms 4				
OW	15.9996	0.0		
HW	1.0008	0.52422		
HW	1.0008	0.52422		
MW	0.0	-1.04844		
constraints 3				
1	2	0.95720000		
1	3	0.95720000		
2	3	1.51390065		
rigid 1				
2	1	4		
finish				
vdW 6				
OW	OW	lj	0.680946	3.16435
OW	HW	lj	0.0	0.0
HW	HW	lj	0.0	0.0
OW	MW	lj	0.0	0.0
HW	MW	lj	0.0	0.0
MW	MW	lj	0.0	0.0
close				

CONTROL NVE

TIP4P model of water

integrator velocity verlet
temperature 300.00
ensemble nve

steps 150000
equilibration 100000
multiple step 1
scale 10
print 50
stack 100
stats 10
rdf 1
traj 1 100 0

timestep 0.001
cutoff 9.50
delr width 0.5000
ewald precision 1d-6
shake tolerance 1.0E-5
quaternion tolerance 1.0E-5
print rdf

job time 300000.00
close time 500.00

finish

CONTROL NPT 300 K 1 atom

TIP4P model of water

integrator velocity verlet
temperature 300.00
ensemble npt hoover 0.1 0.2

steps 150000
equilibration 100000
multiple step 1
scale 10
print 50
stack 100
stats 10
rdf 1
traj 1 100 0

timestep 0.001
cutoff 9.50
delr width 0.5000
ewald precision 1d-6
shake tolerance 1.0E-5
quaternion tolerance 1.0E-5
print rdf
pres 0.001

job time 300000.00
close time 500.00

finish

TIP4P/Ew model: *NVE* ensemble results

movie

temp_tot 3.0270E+02 K r.m.s. fluctn. 6.4702E+00 K

Approximate 3D Diffusion coefficients ($10^{-9} \text{ m}^2 / \text{s}$)

	atom	D
	OW	2.6537E+00
	HW	2.6627E+00
	MW	2.6538E+00

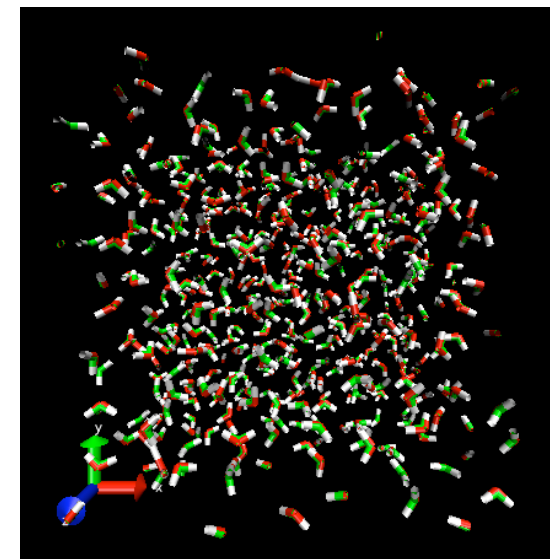
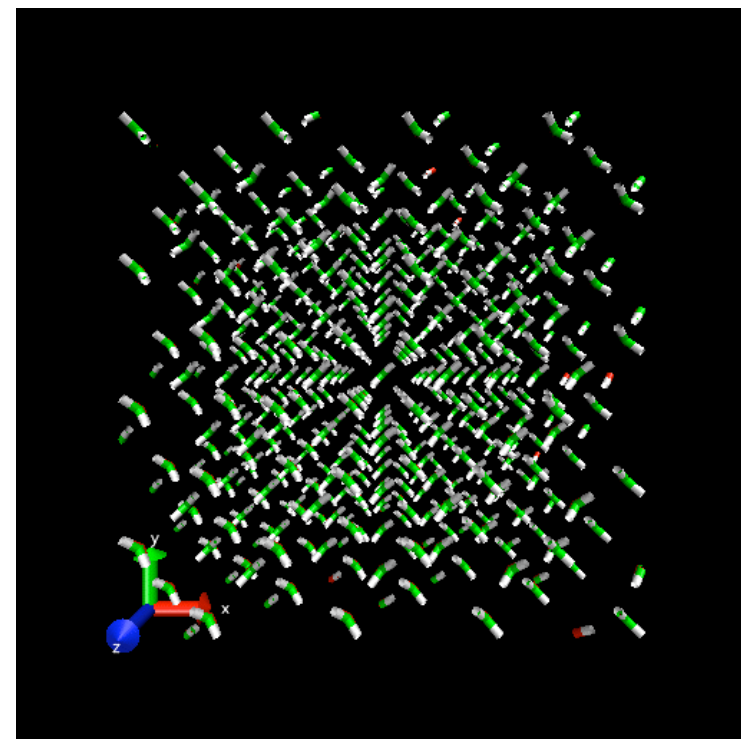
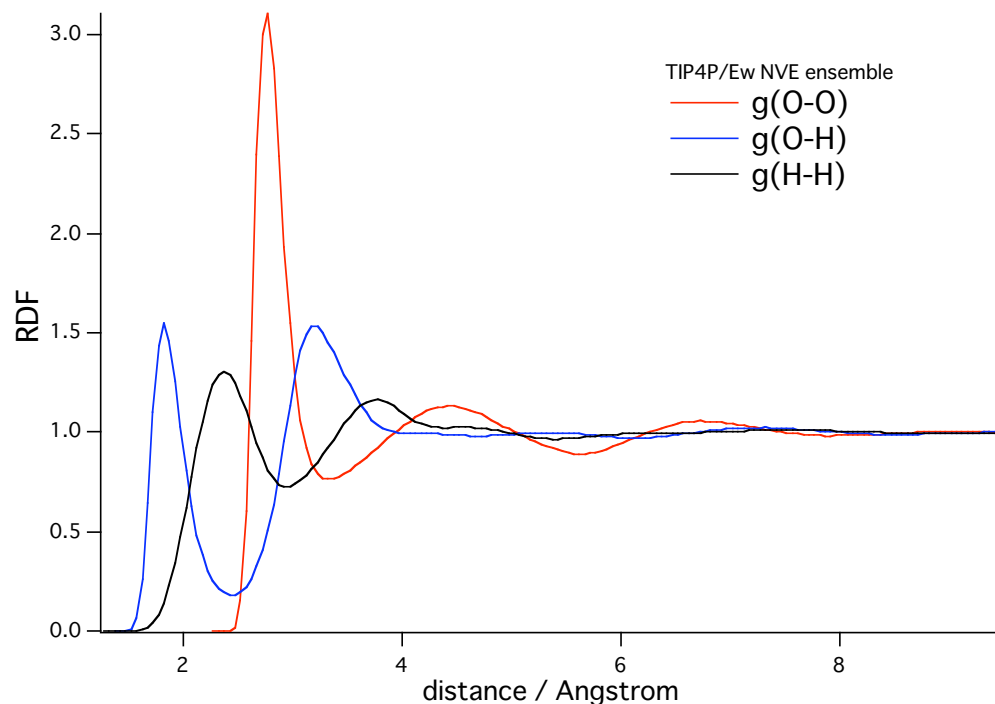
Average pressure tensor

1.8443E-02	6.8694E-03	1.4413E-02
6.8694E-03	3.8647E-02	2.9359E-02
1.4413E-02	2.9359E-02	4.2101E-02

r.m.s. fluctuations

7.6492E-01	5.0866E-01	5.1297E-01
5.0866E-01	7.9529E-01	5.0658E-01
5.1297E-01	5.0658E-01	8.0716E-01

trace/3. 4.3064E-02



time elapsed since job start =

34789.530 seconds

TIP4P/Ew model: *NPT* ensemble results

Temp. 2.9996E+02 K +/- 7.7193E+00

Approximate 3D Diffusion coefficients
(10⁻⁹ m² / s)

atom	D
OW	2.5677E+00
HW	2.5421E+00
MW	2.5601E+00

movie

Average pressure tensor

r.m.s. fluctuations

2.3043E-02	-3.2620E-02	1.5465E-02	8.9626E-01	5.0277E-01	4.8920E-01
-3.2620E-02	4.4084E-03	-1.0378E-02	5.0277E-01	8.5233E-01	4.9858E-01
1.5465E-02	-1.0378E-02	-1.3775E-02	4.8920E-01	4.9858E-01	8.6417E-01

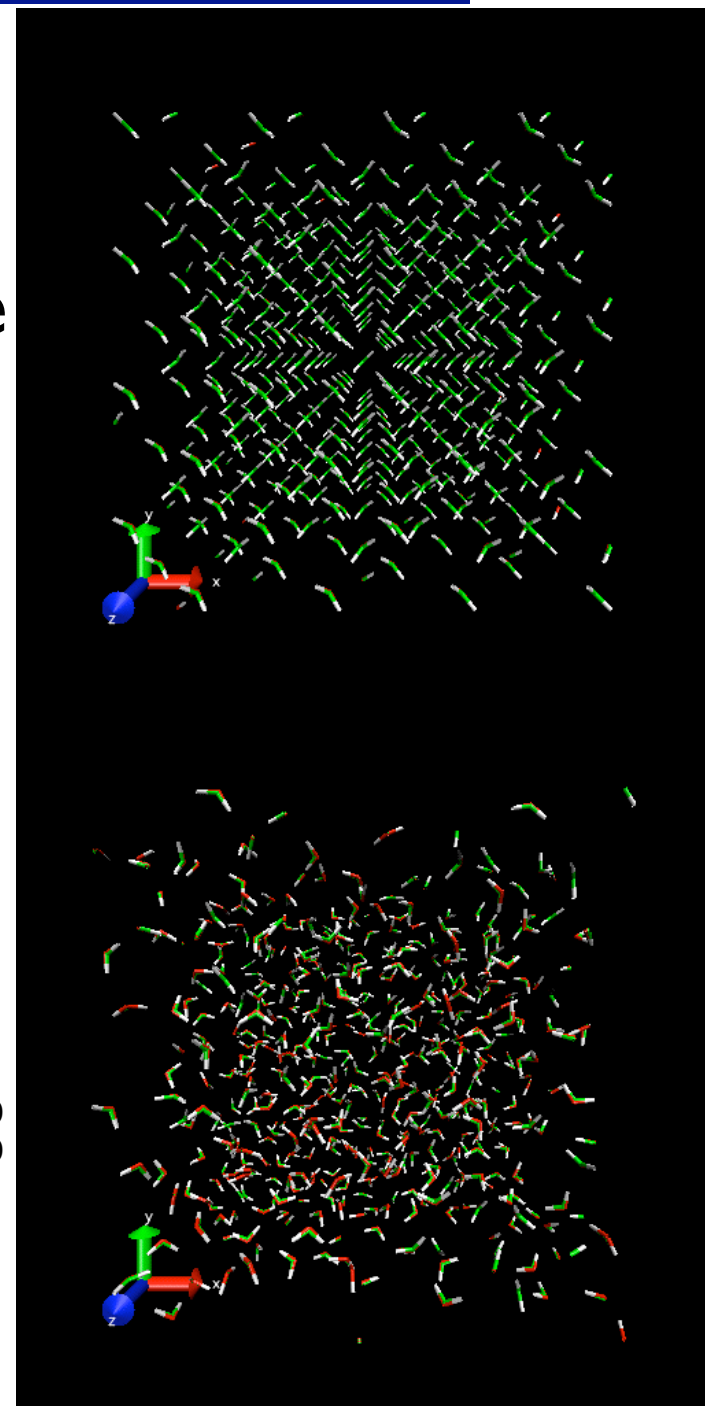
trace/3. 4.5587E-03

Average pressure
4.5587E-03 katom

Average cell vectors

r.m.s. fluctuations

24.8869811588	.0000000000	.0000000000	1.0249E-01	0.0000E+00	0.0000E+00
.0000000000	24.8869811588	.0000000000	0.0000E+00	1.0249E-01	0.0000E+00
.0000000000	.0000000000	24.8869811588	0.0000E+00	0.0000E+00	1.0249E-01



time elapsed since job start = 34773.057 seconds

TIP4P/Ew model: *NPT* ensemble results

cf. exp. results

